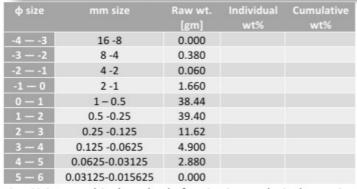


GRAIN SIZE ANALYSIS



 Using graphical method of grain size analysis determine Mode, Mean, Median, Sorting, Skewness, & kurtosis

	-
Mean (M)	$M = \frac{\phi_{16} + \phi_{50} + \phi_{84}}{3}$
Median (Md)	$Md = \Phi_{50}$
Sorting (σ)	$\sigma_{\phi} = \frac{\Phi_{84} - \Phi_{16}}{4} + \frac{\Phi_{95} - \Phi_{5}}{6.6}$
Skewness (SK)	$Sk = \frac{\phi_{16} + \phi_{84} - 2\phi_{50}}{2(\phi_{84} - \phi_{16})} + \frac{\phi_5 + \phi_{95} - 2\phi_{50}}{2(\phi_{95} - \phi_5)}$
Kurtosis	$KG = \frac{\Phi_{95} - \Phi_5}{2.44(\Phi_{75} - \Phi_{25})}$
Mode	$Mo = \frac{\Phi_{max} + \Phi_{min}}{2} of \max wt$
Individual wt%	$wt_{1_x}\% = \frac{Raw\ wt_x\ [g]}{\Sigma raw\ wt\ [g]}$
Comu. wt%	$wt_x\% = wt_{1x} + \Sigma wt_{1(< x)}$

2. Describe the verbal terms of these grain size parameters (use the following tables)

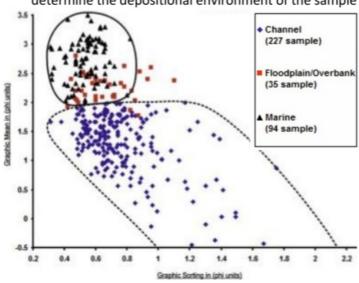
Grain size [Φ]	Terms for sorting
<0.35	Very well sorted
0.35 - 0.50	Well sorted
0.50 - 0.71	Moderately well sorted
0.71 – 1.00	Moderately sorted
1.00 - 2.00	Poorly sorted
>2.00	Very poorly sorted

Sk	Terms for skewness				
<-0.3	Strongly coarse-skewed				
-0.100.30	Coarse-skewed				
+0.100.10	Near-symmetrical				
+0.30 - +0.10	Fine-skewed				
>+0.30	Strongly fine-skewed				

- -ve skewed: If the distribution has a coarse 'tail' (there is an excess of coarse material)
- +ve skewed: if the sediment has a fine 'tail'
- No skew: If the distribution is symmetrical

Length (mm)	**	Class	Sediment/ rock name
— 4096 ————————————————————————————————————		block	mega- conglomerate
— 4096 —— -1 — 2048 —— -1	l vc	• .	
— 1024 ————————————————————————————————————	10	boulder	
— 512 — -	m		
— 256 — -	1 f 1		
— 128 ——	c	cobble	gravel conglomerate
_ 64	l f	coppie	congiomerate
_ 32	l vc l		,
16	c	pebble	40 4 7 9
81	m	peoble	
_ 4 _ =	l f	3 y 4	·
_ 2		granule	* 1, * .
	vc.		
	c		sand
	m	sand	sandstone
0.23	f		~
0.125	vf		2 . 7
	c		
	m	silt	silt
	f	3111	siltstone
	vf.		
2 Using the m		clay	clay claystone

Using the mean-sorting diagram of Amireh (2015) determine the depositional environment of the sample



1st step: calculate individual & cumulative wt%

Individual wt% (wt₁%)

$$wt_{1_x}\% = \frac{Raw\ wt_x\ [g]}{\Sigma raw\ wt\ [g]}$$
0.380g/99.34g = 0.382%
0.060g/99.34g = 0.060%
. . . .
2.880g/99.34g = 2.899%

Cumulative wt%

$$wt_x\% = wt_{1x} + \Sigma wt_{1(
 $0.382\% + 0 = 0.382\%$
 $0.060\% + (0.382\%) = 0.442\%$
 $1.671\% + (0.442) = 2.113\%$$$

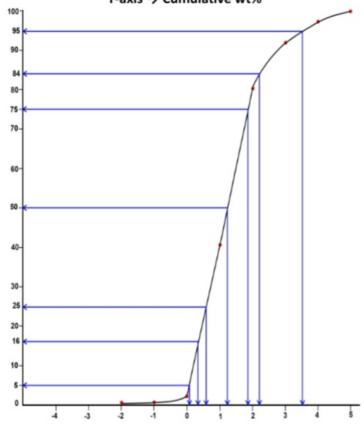
. . .

2.899% + (97.11%) = 100.0%

ф size	mm size	Raw wt. [gm]	Individual wt%	Cumulative wt%
-4 — -3	16 -8	_	_	-
-3 — -2	8 -4	0.380g	0.382%	0.382%
-2 — -1	4 -2	0.060g	0.060%	0.442%
-1 — 0	2 -1	1.660g	1.671%	2.113%
0 — 1	1-0.5	38.44g	38.70%	40.81%
1 — 2	0.5 -0.25	39.40g	39.66%	80.47%
2 — 3	0.25 -0.125	11.62g	11.70%	92.17%
3 — 4	0.125 -0.0625	4.900g	4.933%	97.11%
4 — 5	0.0625-0.03125	2.880g	2.899%	100.0%
5 — 6	0.03125-0.015625	-	-	-
	Total	99.34g	100.0%	

2nd step: plot the cumulative frequency curve

X-axis → Grain Size (Φ-unit) Y-axis → Cumulative wt%



3rd step: Calculate parameters & Describe verbal terms

$$\Phi_5 = 0.06$$
, $\Phi_{16} = 0.32$, $\Phi_{25} = 0.6$, $\Phi_{50} = 1.25$
 $\Phi_{75} = 1.86$, $\Phi_{84} = 2.18$, $\Phi_{95} = 3.55$

Mode

$$\mathbf{M} = \frac{\boldsymbol{\phi}_{max} + \boldsymbol{\phi}_{min}}{2}_{max \ wt} = \frac{2+1}{2}_{39.4g} = 1.5 \Phi$$

Mean

$$\mathbf{M} = \frac{\boldsymbol{\phi_{16}} + \boldsymbol{\phi_{50}} + \boldsymbol{\phi_{84}}}{3} = \frac{0.32 + 1.25 + 2.18}{3} = 1.25\Phi$$

Median = Φ₅₀ = 1.25Φ

Sorting

$$\sigma_{\phi} = \frac{\Phi_{84} + \Phi_{16}}{4} + \frac{\Phi_{95} - \Phi_{5}}{6.6}$$
$$= \frac{0.32\Phi + 2.18\Phi}{4} + \frac{3.55\Phi - 0.06\Phi}{6.6} = 1.17\Phi$$

Skewness

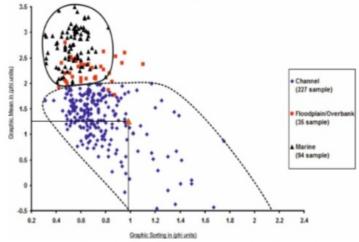
$$Sk = \frac{\boldsymbol{\phi_{16}} + \boldsymbol{\phi_{84}} - 2\boldsymbol{\phi_{50}}}{2(\boldsymbol{\phi_{84}} - \boldsymbol{\phi_{16}})} + \frac{\boldsymbol{\phi_{5}} + \boldsymbol{\phi_{95}} - 2\boldsymbol{\phi_{50}}}{2(\boldsymbol{\phi_{95}} - \boldsymbol{\phi_{5}})}$$
$$= \frac{0.32 + 2.18 - 2x1.25}{2(2.18 - 0.32)} + \frac{0.06 + 3.55 - 2x1.25}{2(3.55 - 0.06)} = 0.16$$

Kurtosis

$$KG = \frac{\boldsymbol{\phi_{95}} - \boldsymbol{\phi_5}}{2.44(\boldsymbol{\phi_{75}} - \boldsymbol{\phi_{25}})} = \frac{3.55 - 0.06}{2.44(1.86 - 0.6)} = 1.14$$

Parameters	Value	Verbal terms
Mean (M)	1.25Ф	Medium Sand
Median (Md)	1.25Ф	Medium Sand
Mode	1.50Ф	Medium Sand
Sorting (σ)	1.17Φ	Poorly sorted
Skewness (SK)	0.16Ф	Fine-skewed
Kurtosis	1.14Φ	

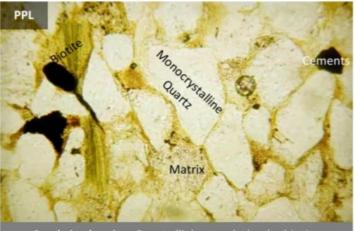
4th step: determine the depositional environment



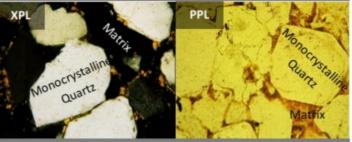
The dispositional environment is Channel

Sandstones Mineral Composition

- The clastic texture consists of:
 - Framework components: clasts, grains, or particles (include quartz, feldspar, & rock fragments)
 - 2. Matrix: grains < silt (<0.63mm) between the clasts
 - 3. Cement: filling pore spaces between grains & matrix
- 4. Pore spaces: empty (not filled by cement or matrix) الفرق بين ال cement وال matrix هو اننا نستطيع تميز ال matrix لإنها تتميز بخصانص بصرية محددة ولكن لا نستطيع تميز ال matrix



Sand-sized grains: Quartz (light or colorless) + biotite ثشكل هذه البلورات ال framework لانها متصلة مع بعضها البعض Matrix : (<63µm) yellow لا يمكن تعييز خصائصها Cements : black grains



البلورات الصفراء هي كوارتز وتم تعييزها بواسطة ال polyhedral) والذي ينتج بسبب ال overgrowth

ال overgrowth هو جزء من البلورة يحيط بجزء اخر (يسمى core) ويفصل بينهما خط (يسمى dust line) والذي تبلور ببيئة مختلفة عن ال core لا يوجد cement لان المادة الصغيرة بين البلورات لا يمكن تمييز خصائصها البصرية وبالتالي هي matrix

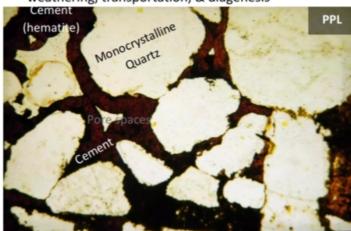
Quartz [Qz, SiO₂]

- The most common detrital mineral in all sandstones
- No sandstone without Qz, because Qz most stable silicate light mineral under sedimentary conditions
- Monocrystalline Qz: consists of single crystal بلورة واحدة
- Polycrystalline Qz: each grain consists of ≥ 2 crystals
- Non-undulose Qz: having a straight or unit extinction under cross polarized light عند التدوير
- Undulose: having wavy or undulose extinction
 تطفئ من طرف البلورة للطرف الاخر مرورا بالوسط اثناء الدوران

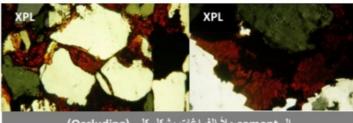
- Qz characterized by presence of some inclusions such as needles of sillaminite, vacuoles of fluids or minute crystals (tourmaline, mica or rutile)
- Qz parent rock: plutonic rocks, acid gneisses, schist, & in some cases from pre-existing sandstones

Properties of Qz that can be employed to infer its source rock				
Sources	Quartz properties			
Volcanic rock	Non-undulose Monocrystalline			
	no inclusions & euhedral crystals			
Hydrothermal vein	Have fluid-filled vacuoles			
Metamorphic rocks	Poly., Elongate, preferred orientation			
Plutonic rocks	Large with Undulose extinction			
Older Sandstone	Non-undulose mono., overgrowth			

 Non-undulose monocrystalline is the most common in most sandstones due to its higher stability during weathering, transportation, & diagenesis



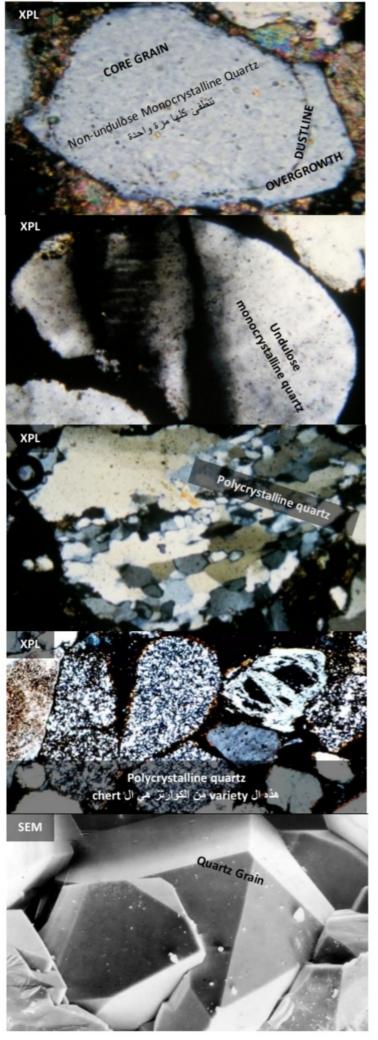
Cement: red or brown which filled pore spaces (hematite) هو cement لان خصائصه البصرية واضحية ويمكن معرفة انه اكميد الحديد الاماكن الموجودة بين السمنت هي فر اغات لم تملأ ونستطيع التأكد من ذلك بادخال الضوء المستقطب حيث يصبح لونها اسود (isotropic) ولونها مثل الكواترز color less لانها مملوءة بالصمغ الذي تحضر به ال

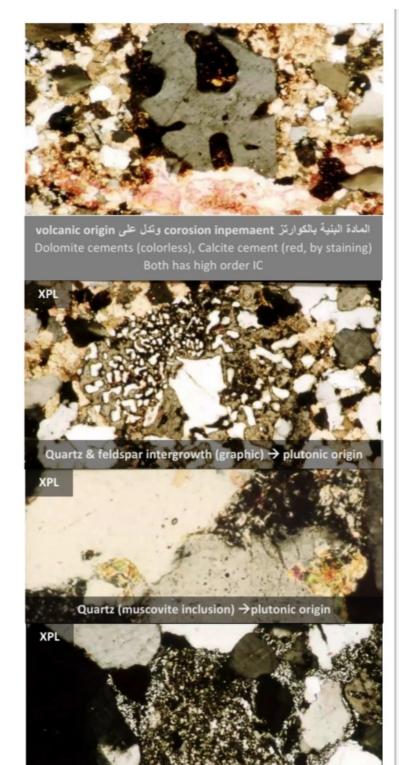


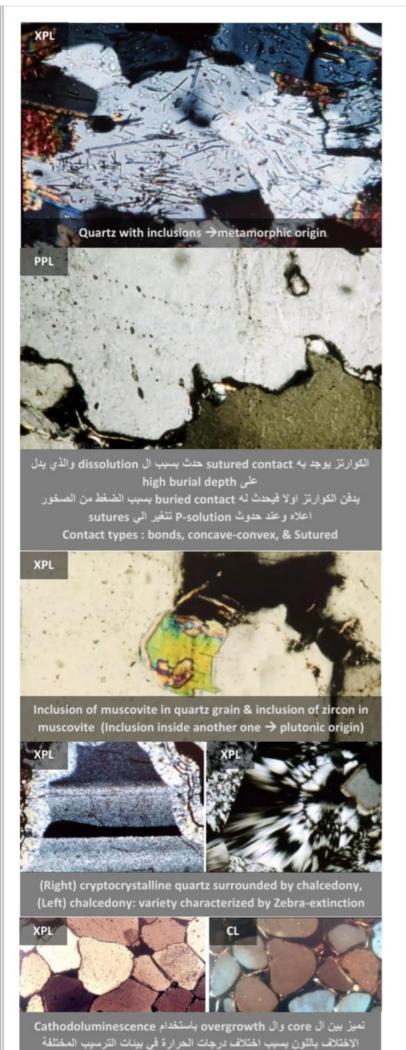
ال cement ملأ الفراغات بشكل كلي (Occluding) XPL PPL

Quartz + Glauconite (green) + Muscovite (replacement by Glauconite in part "alteration")

Cements: Fe-dolomite "ankerite" (cleavage, & high order IC)

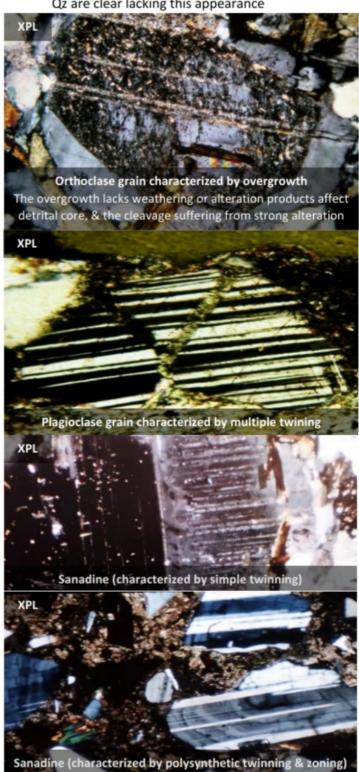


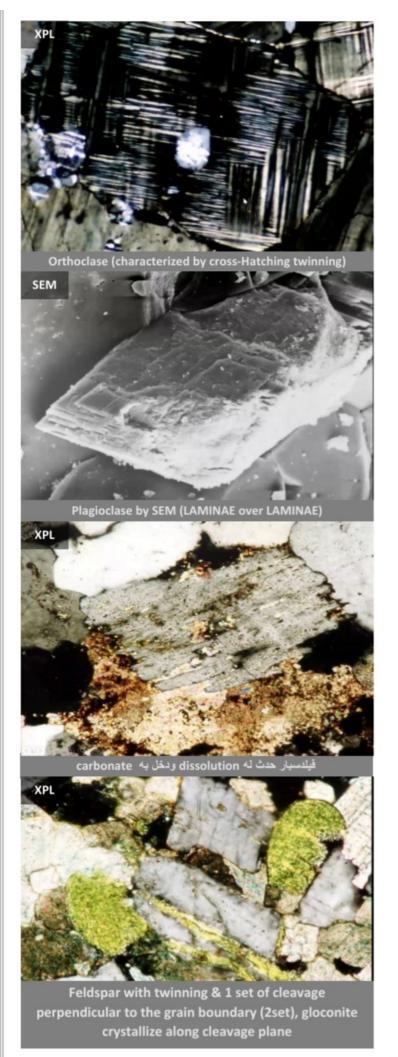




Feldspars [Fs, (Na,K)AlSi₃O₈, CaAl₂Si₂O₈]

- Less common in sandstones than quartz
- The reason for lower concentration than quartz:
 - 1. lower chemical stability to chemical weathering
 - 2. lower resistance against mechanical abrasion due to well-developed cleavage
- Feldspar grains Vs quartz grains in thin sections
 - Twining: Cross-hatch twining of microcline, & Carlsbad twining of orthoclase
 - Cleavage: in Fs, chemical alteration products concentred along cleavage planes
 - Appearance: chemical weathering of Fs imparts a turbid color, cloudy, or dusty appearance, whereas Qz are clear lacking this appearance







ببيئة بعيدة عن التجوية, والبلورة في الوسط (core) هي detrital لانها rounded & include weathering product



سبار حدث له internal dissolution واللون الاخضر هو ص نتعرف على ال pore spaces

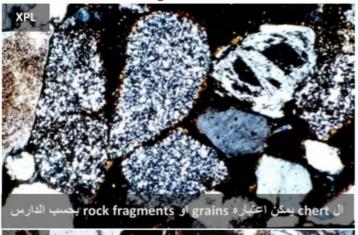
- * Type of pore spaces:
- داخل البلورة Intrapore
- ين الحبات 2. Extrapore

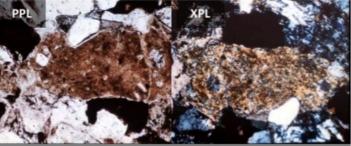


Feldspar core with weathering product along cleavage plane, & dust line divided the core from overgrowth

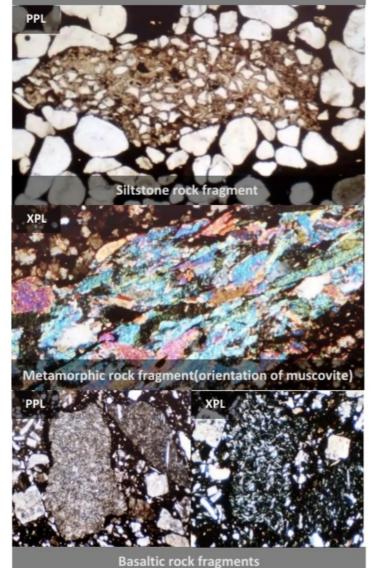
Rock Fragments [lithic fragments]

more abundant in conglomerate than sandstones





Mudstone rock fragment تتفتت تحت الضغط، وهي ليست matrix لان لها boundary فی XPL بدأنا نری IC for clay minerals ما یدل انها XPL





augite crystal "ophitic texture")

Other Components

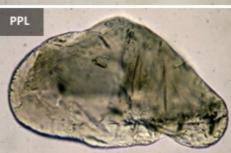
- Detrital micas: biotite, muscovite, & rarely chlorite
 - Form flakes & arranged parallel to bedding plane
 - muscovite is more common than biotite (unstable)
 - If biotite is present, it mainly altered into Fe-oxide, chlorite, or clay mineral (illite or kaolinite)
- Carbonate fragments: shells, fossils, ooids, & intraclasts
- Galuconite & phosphatic grains
- Heavy minerals (HM): with a concentration < 1%, from separation from the light minerals (Qz & Fs)
 - > Zircon, Tourmaline, & Rutile (stable to ultrastable)
 - Apatite, Garnet, Epidote, Sillimanite (metastable)
 - Olivine, Pyroxene, & Hornblende (Unstable)

Zircon: very high relief, high IC, inclusions, the borders of the grain are heavily abraded



Tourmaline: strong relief, high order IC, &

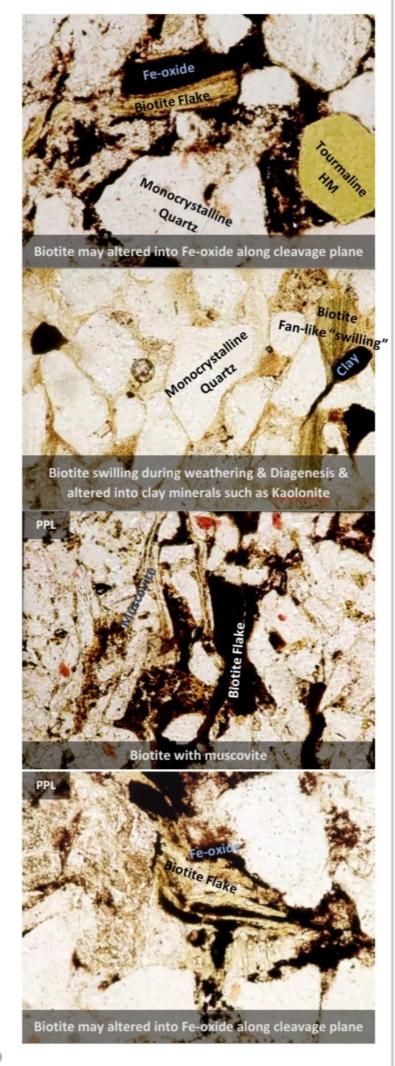
strong pleochroism

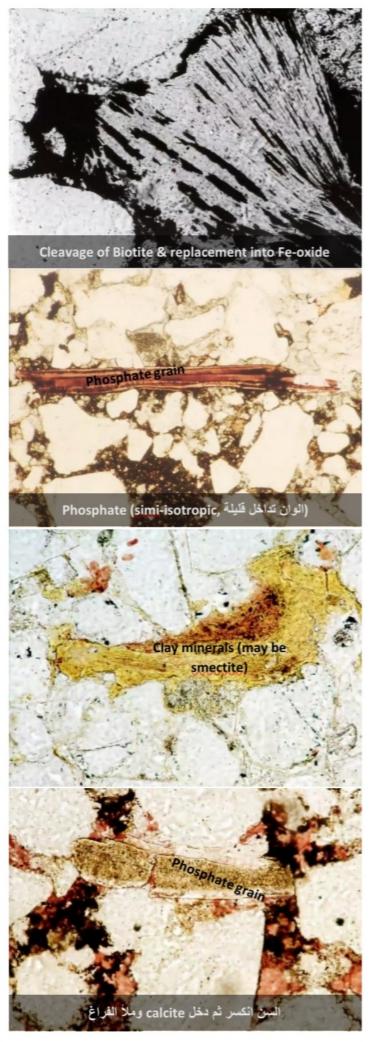


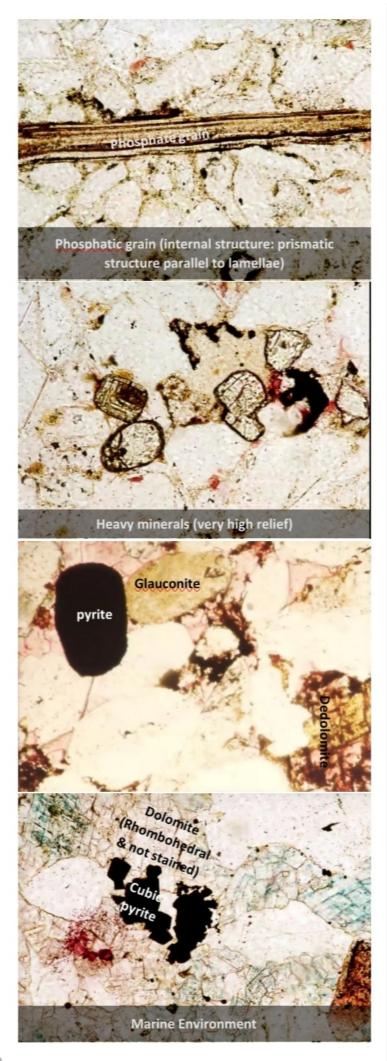
Rutile: red, very strong relief & IC, masked by the original color of the mineral











Diagenesis of Clastic Sedimentary Rocks

- Diagenesis: all physical & chemical processes that affect the sediments from sedimentation until the on set of low grade metamorphism
- Factors that Control diagenesis: Provenance, Burial Depth, Depositional Environment, Tectonism & Thermal Activities, Uplift & Subaerial Weathering

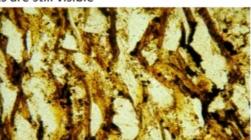
Physical or Mechanical Diagenesis

 Results from the weight of overlying sediments & starts by mechanical compaction followed by pressure solution due to increasing burial depth

At a low burden depth (early compaction)

- elongate (or plate) grains (e.g. mica flake) respond to P by orientation of their long axis (maximum surfacearea) parallel to bedding planes
- Point contacts are still visible

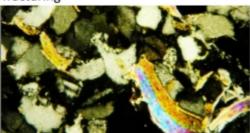
Quartz, & flaky micas (biotite & muscovite grains which arranged parallel to bedding plane)



By increasing burial depth (pseudoplastic deformation)

- ductile grains (mica flakes) respond by pseudoplastic deformation (pseudoplastic bending for ductile grains)
- brittle grains fracturing

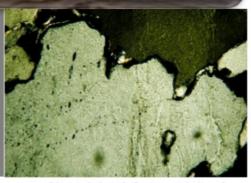
Pseudoplastic bending for ductile grains Biotit (masking to IC) & Muscovite (2nd order)





Increasing burial depth (1000-1500, Pressure Solution)

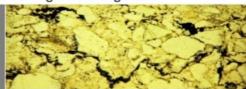
 Qz at point contacts subjected to high effective P, causes preferential solution (Qz characterized by concavo-convex & sutured contacts) 2 quartz grains suffering from Psolution & characterized by sutured contact 3 and contact formed after burial depth & as increase depth become concaveconvex & finally suture



By subsequent increase of burial depth (Microsylolites)

 Microsylolites: zigzag cross-cut quartz grains, marked by concentration of insoluble material that resulted from P-solution & migrated along them

Microstylolite in Qz Devolved after suture contact & represents greater burial depth



Summary

machenical diagenesis depending on burial depth, as burial depth increases: compaction → pseudoplastic deformation of ductile grains & fractured of brittle grains → P-Solution (Bond → concave-convex → suture → Microsylolites)

Chemical diagenesis

- Silica cement: precipitated around the quartz grains in form of a quartz overgrowt
 - > Qz overgrowth: optical continuity with detrital core
 - Qz overgrowth characterized by presence of a dust line which is a coat of Fe-oxides or clay minerals
 - If there is no dust line, it is difficult to distinguish between the quartz overgrowth and the core & a cothodoluminescence study is required
 - Sentaxial overgrowth: has the same chemical composition (& optical properties) as detrital core
 - Epitaxial overgrowth: has the different chemical composition (optical properties) from detrital core

Sentaxial Overgrowth has a distinctive crystal phases لازه لم يحدث له عمليات تجوية وتعرية وترسب في diagenesis



تلتحم ال ل للبلورات المختلفة مع بعضها لبعض لتنتج mosaic texture



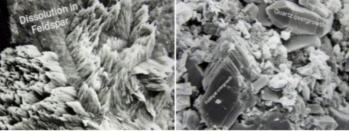


form of feldspar overgrowth around detrital core

Overgrowth is clear & have sharp crystal phases



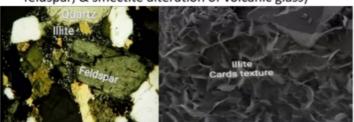
tinuity with detrital core (epitaxial)

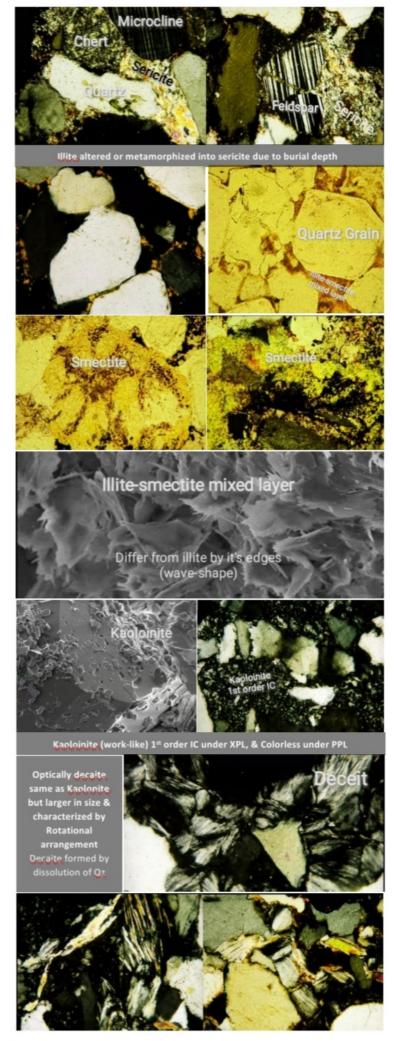


in development of

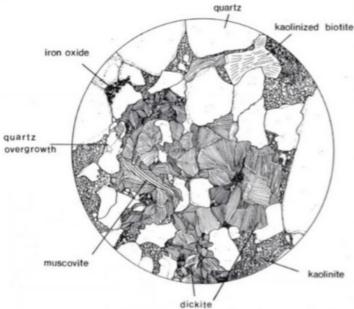


Clay mineral occur as pore-filling cement or caly rims around detrital grains produced by replacement or alteration of detrital minerals (e.g. kaolinite replacing feldspar, & smectite alteration of volcanic glass)

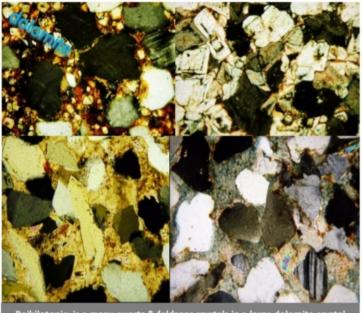






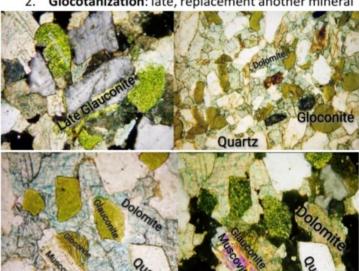


- Carbonate cementation calcite or dolomite
 - Calcite: poikilotopic cement, consists of large patches of calcite crystals engulfing sand grains
 - drusy calcite equant calcite crystals that fill the pore spaces & show an increase in crystal size towards the center of the original cavity
 - Dolomite pore-filling microcrystalline rhombs to coarse anhedral mosaics & large poikilotopic
 - > Fe-dolomite (Fe-rich): under reducing conditions
 - Dolomite formed from micrite in marine setting



Poikilotopic: is a many quartz & feldspar crystals in a large dolomite crystal

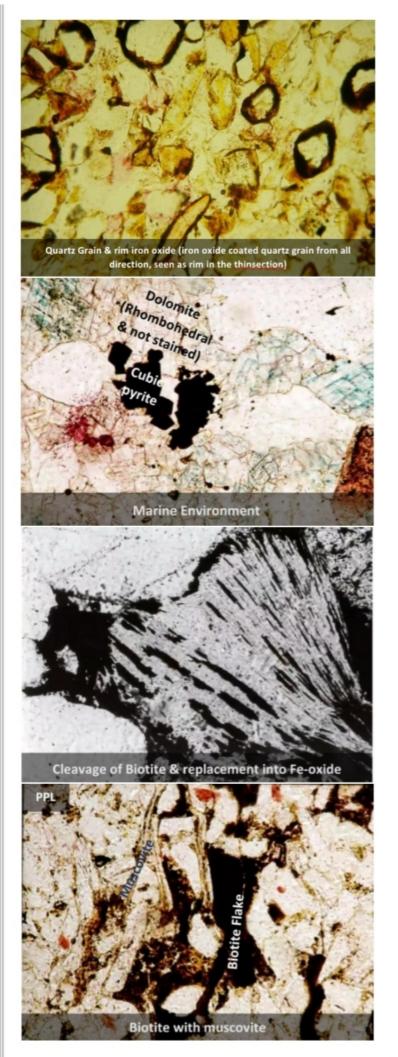
- Glauconite: formed during 2 stages
 - Sensedimantary stage: is an Early diagenetic processes (precipitated during precipitation)
 - 2. Glocotanization: late, replacement another mineral

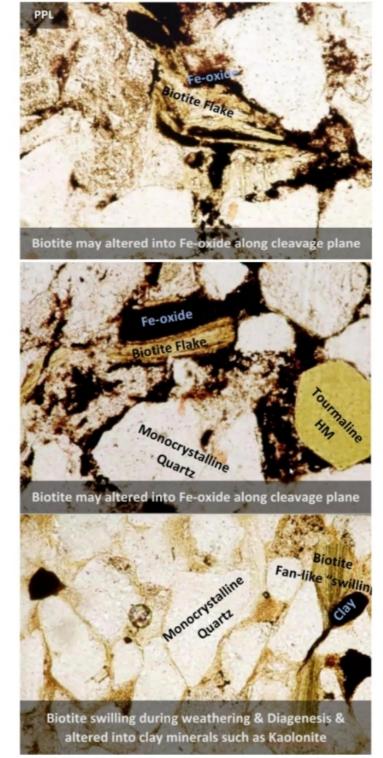


FeldsparAnataze



- Fe-oxide: red color (hematite), Occurs as pore lining or filling, & rim coating detrital grains, stain authigenic clay minerals & feldspar grains along fractures or cleavage
 - Hematite replaces variably biotite flakes along cleavage planes, ranging from scattered spots or lamellae to complete replacement of flake, which is identified as a former detrital biotite flake on base of a fan- or flake-like shape
 - Hematite "optical": red; translucent; & filling pore space, dark brown; opaque; as rim cement
 - Source of Fe-oxides cements:
 - detrital origin (amorphous Fe-compounds) via humid tropical weathering, transported & deposite, & then converted (aged) to hematite
 - 2. **yellow-brown coats on detrital sand grains** through chemical weathering & converted into hematite after deposition (ageing)
 - pure diagenetic from intrastratal solution of detrital ferromagnesian silicates, If the diagenetic environment is oxidizing the Fe precipitated as hematite or hydrated Fe-oxide precursor (converts to hematite on ageing)





- Diagenesis of Jordanian sandstone succession from the Lower Cambrian to the Upper Cretaceous is influenced by depositional environment, burial depth & uplift
 - Most common diagenitic processes Qz-overgrowth, kaolinite authigenesis, & Fe-oxide cementation, & these processes are depositional environment & burial depth *irrelevant*
 - Marine sandstones exhibit authigenesis of feldspar, glauconite, dolonite, smectite or illite-smectite mixed Byer, & pyrite
 - Burial depth controlled Qz P-Solution & subsequent microstylolite development, alteration of ilite into sericite, dickite formation, & tourmaline, Corrosion
 - uplift governed Fe-oxide cementation

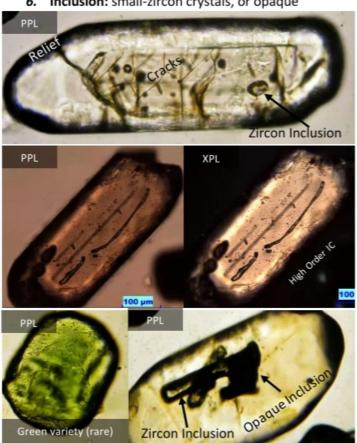
Heavy Minerals

ULTRASTABLE

- All heavy minerals characterized by high relief due to high refractive index relative to epoxy
- Found in all sandstones as accessory minerals (<1%Vol)
- Separated from another minerals in sandstones using heavy liquid because all have specific gravity > 2.9, will quartz & feldspar around 2.5

ZIRCON

- Most stable heavy mineral, found in all sandstones due to high resistance to machanical & chemical weathering during transportation
- Optically:
 - 1. Very High Relief: due to high Refractive index, higher than all heavy minerals
 - 2. Color: (PPL) Colorless (most common), or Green (rare), (XPL) Very High Order IC
 - 3. Lack of pleochroism (لا يتغير لونه عند تدوير الستيج)
 - 4. Rounding grains (Lack of edges): due to long distance of transportation
 - 5. Presence of Cracks: due to decay of radioactive elements in the zircon grains
 - 6. Inclusion: small-zircon crystals, or opaque



RUTILE (Ti-Oxide)

- Optically:
 - 1. Slightly pleochroism
 - 2. High relief (less than zircon)
 - 3. Presence of inclusions
 - 4. Color: (PPL) Bloody red (most common), golden or yellow variety, (XPL) masks IC
 - 5. Shape: Twinning, Rounded, may take hart-shape



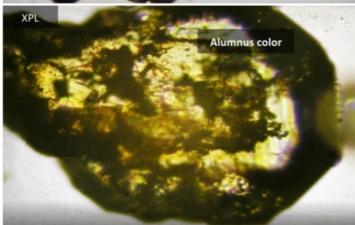
This planes are twinning planes, not cleavage planes because these grains are rounded (transported long distance), ao if this planes are cleavage this grains must be breaking down into smaller minerals along cleavage plane, another evidence arises from pleochroism

ANATASE (Ti-Oxide)

Optically

- 1. Slightly Transparent to Translucent at the edges (colorless), & opaque in the center (black, or brown)
- 2. Very High Relief (high refractive index)
- 3. XPL: 1st order blue (called Alumanuse IC)



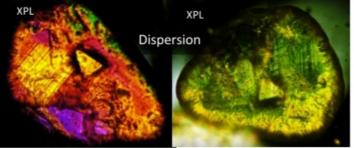


Prookite (Ti-Oxide)

Optically

- 1. High relief
- 2. Color: (PPL) yellow, (XPL) Dispersion between 1st & 2nd IC (rapid change in IC with rotating stage)

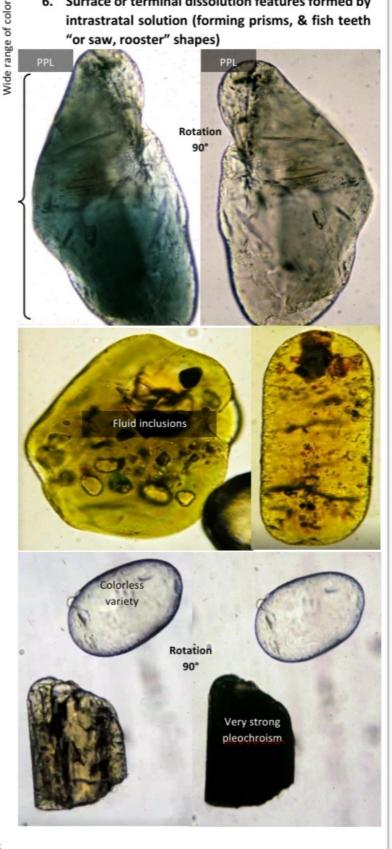




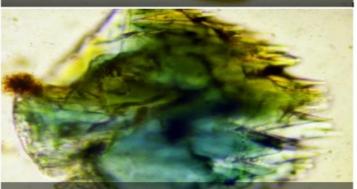
TOURMALINE

Optically

- 1. Color: (PPL) wid range of color in the same grain, or colorless (rare), (XPL) high ordar IC (the colorless variety distinguished from apatite by IC)
- 2. Very strong pleochroism under poth PPL & XPL
- High relief (less then zircon & rutile)
- Presence of inclusions such as fluid inclusions
- Rounded:by sedimentation cycles or transportation
- 6. Surface or terminal dissolution features formed by intrastratal solution (forming prisms, & fish teeth "or saw, rooster" shapes)



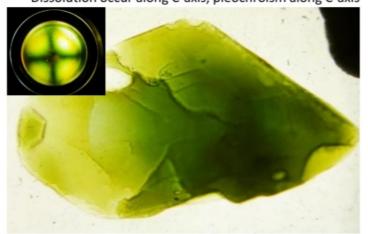
Surface-dissolution features formed by intrastratal solution (prisms)



Terminal-dissolution features formed by intrastratal solution (fish teeth, saw, or rooster shapes)



Dissolution occur along C-axis, pleochroism along C-axis

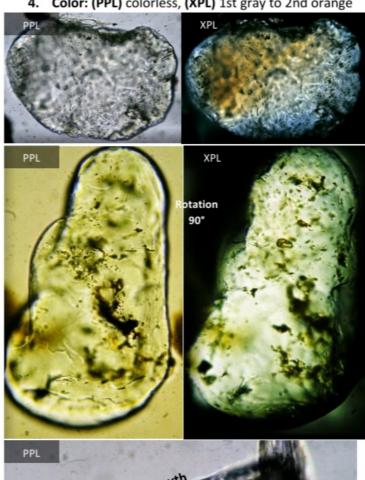


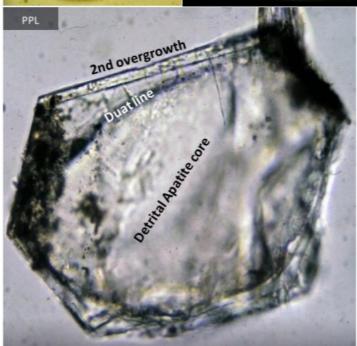
This grain is tourmaline because optical figure show that, this grain is uniaxial grain, not hornblende because hornblende is biaxial

METASTABLE

APATITE

- Optically
 - 1. High relief: has higher refractive index than epoxy
 - 2. Presence of inclusions, dissolution feature, or apatite overgrowth (due to diagenesis)
 - 3. Rounded: sedimentation cycles of transportation
 - Color: (PPL) colorless, (XPL) 1st gray to 2nd orange



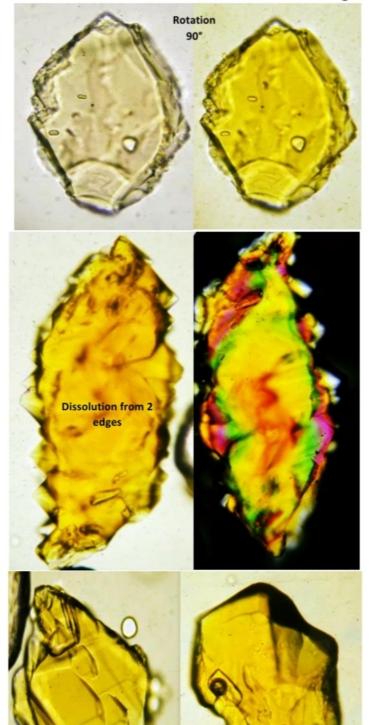


There're 2 stages of overgrowth (not 2 cycles because the first one are very angular, if there's another sedimentation cycle this edges must be rounded)

STAUROLITE

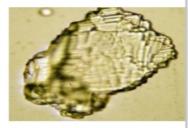
Optically

- 1. Pleochroism, & high relief
- 2. Angular grains due to interstartal solutions
- 3. Presence of inclusions, & overgrowth
- Color: (PPL) colorless, yellow, or yellowish-orange, (XPL) 3rd order interference color
- 5. Terminal surface or terminal dissolution of 2 edges



GARNET

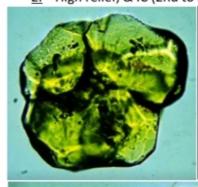
 Optically: Colorless, With surface or terminal dissolution, extinction under XPL (isotropic)

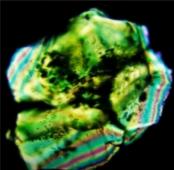


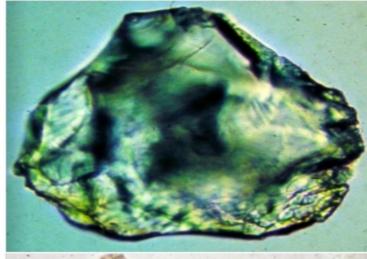
EPIDOTE

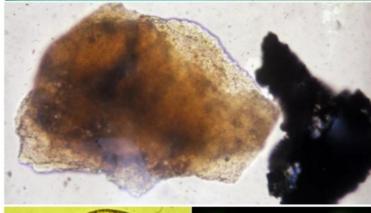
Optically:

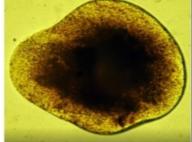
- Yellowsh-Green (Stachia), Colorless (Zeozite),
 Orange (by senziratization during diagenesis)
- 2. High relief, & IC (2nd to 3rd & up to 4th order)

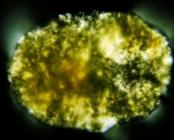




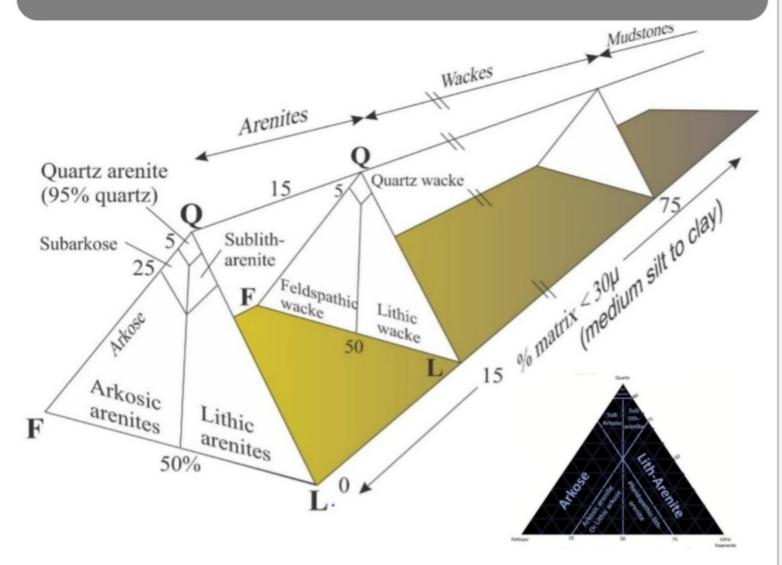








Clastic Sedimentary Rock Classifications



		Fram	ewo	rk Gr	ains					C	ement	S		Р
	Qz		ı	Fs		Fr		Mt	Qz	С	ar	Fe	CI	S
NU	U	Р	K	PI	٧	M	S		Ov	Cal	Dol	He	СІ	3
110		÷	- 11		_						20.	110		

Prefix: NU: Non-undulose, U:Undulose, P: Polycrystalline, K: K-feldspar, PI: Plagioclase, V: Volcanic, M: Metamorphic, S: sedimentary, Mt: Matrix, Qz: Quartz, Ov: Overgrowth, Cal: Calcite, Dol: Dolomite, He: Hematite, CI: Clay Minerals, PS: Pore Spaces, Fs: Feldspar, Fr: Rock Fragments, Car: Carbonates, Fe: Fe-Oxides

1. نحسب مجموع ال grains (S), مجموع ال grains باستثناء

(Y) Framework Grains ومجموع ال (X) pore spaces

$$Y = \Sigma Qz + \Sigma Fs + \Sigma Fr$$
$$X = S - PS$$

2. نحسب ال %Matrix بقسمة ال Ma على مجموع ال

$$Mt\% = \frac{Mt}{X}x100\%$$

3. نحسب ال Modal Qz, Modal Fs, Modal Fragments

$$Modal(Grains) = \frac{\Sigma Grains}{\gamma} x 100\%$$

4. نحسب ال Porosity

$$Porosity = \frac{PS}{S} x 100\%$$

Fs		E.			-	_				
				Mt	Qz	С	ar	Fe	- C	PS
K PI	٧	M	S		Ov	Cal	Dol	Не	CI	
40 10	7	4	-	30	20	18	5	25	7	15
			40 10 7 4	40 10 7 4 -		40 10 7 4 - 30 20	40 10 7 4 - 30 20 18	40 10 7 4 - 30 20 18 5	40 10 7 4 - 30 20 18 5 25	40 10 7 4 - 30 20 18 5 25 7

X = 385

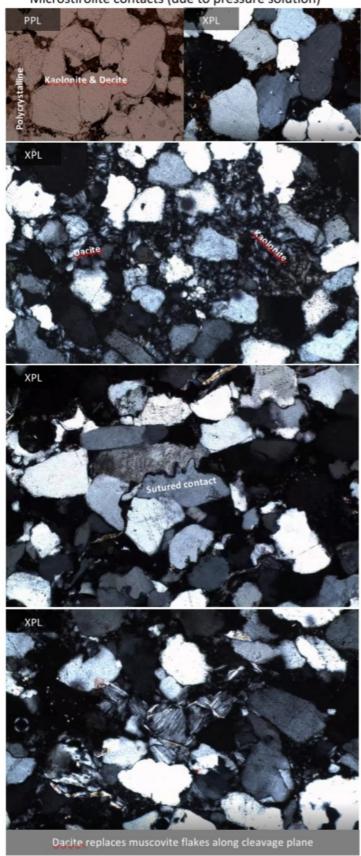
$$Mt\% = rac{30}{385}x100\% = 7.8\%
ightarrow Arenite$$
 $Modal\ Qz = rac{159 + 45 + 15}{280}x100\%
ightarrow 78\%$
 $Modal\ Fs = rac{40 + 10}{280}x100\%
ightarrow 18\%$
 $Modal\ Fr = rac{7 + 4}{280}x100\%
ightarrow 4.0\%$

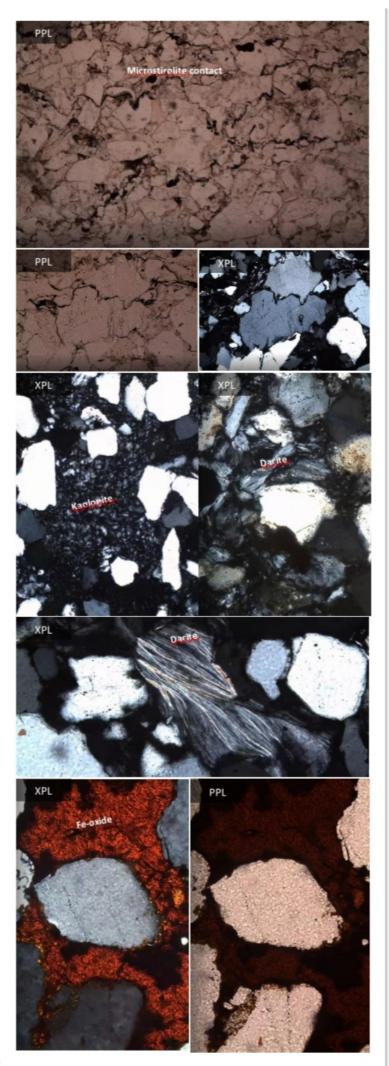
 $4.0\%Fr + 18\%Fs + 78\%Qz = 100\% \rightarrow Subarkose$ The classification of this rock : Calcite-cemented, Hematitic-Subarkosic-Arinite Sandstone

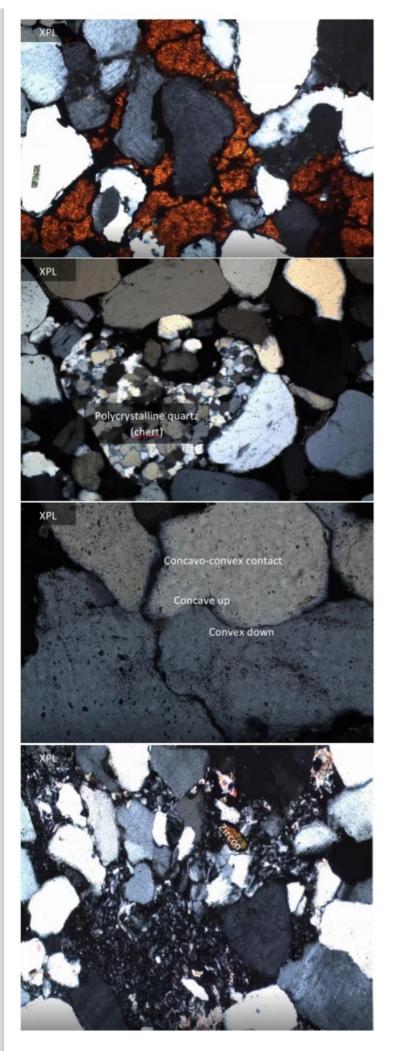
من الصور قم بعد كل الحبات الموجودة ثم اقسم عدد ال Fs على Qz وحدد
 كمية ال Matrix وستحصل على اسم الصخر

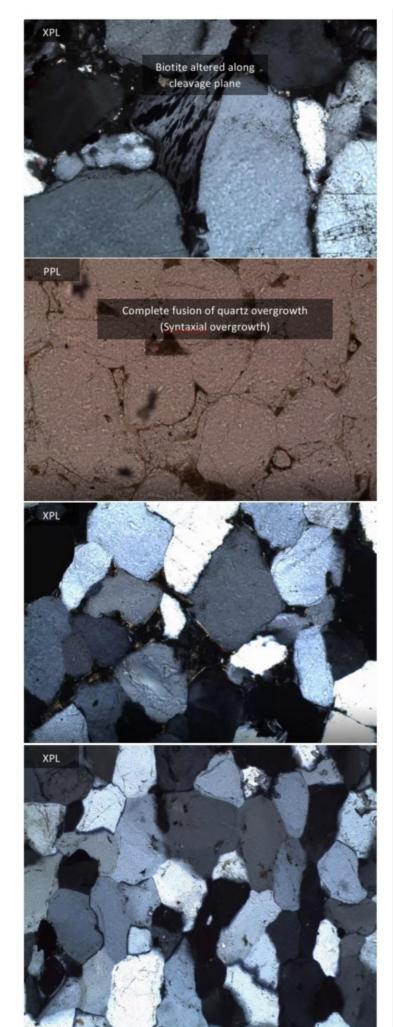
ARENITE SANDSTONES

- Composition: Mostly Undulose & Non-Undulose Monocrystalline Quartz & Polycrystalline Quartz
- Cements: Quartz Overgrowth, Clay Minerals such as Kaolonite, & Decite (Formed by alteration of mica flakes along cleavage planes) or Fe-oxide (Hematite)
- Characterized by sutured contact, Concave-Convex, or Microstirolite contacts (due to pressure solution)

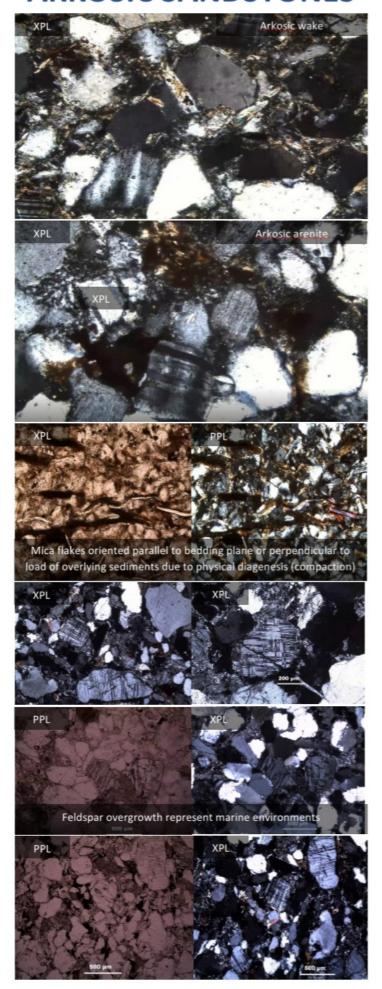


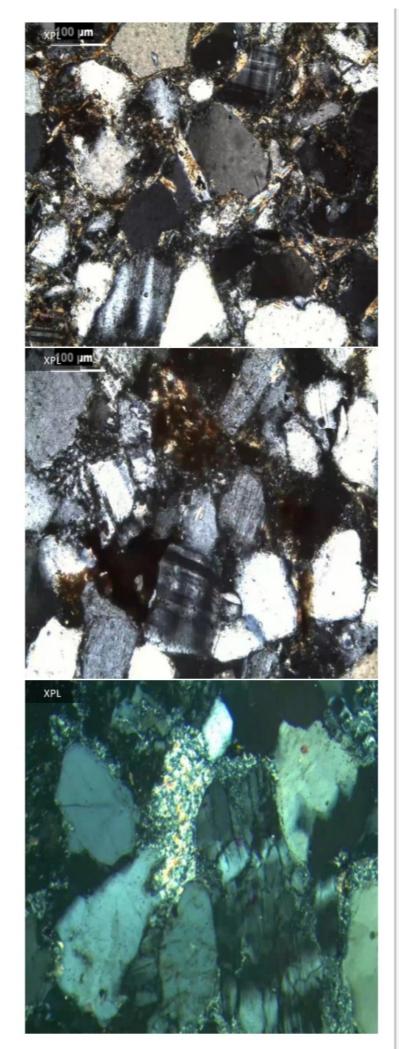




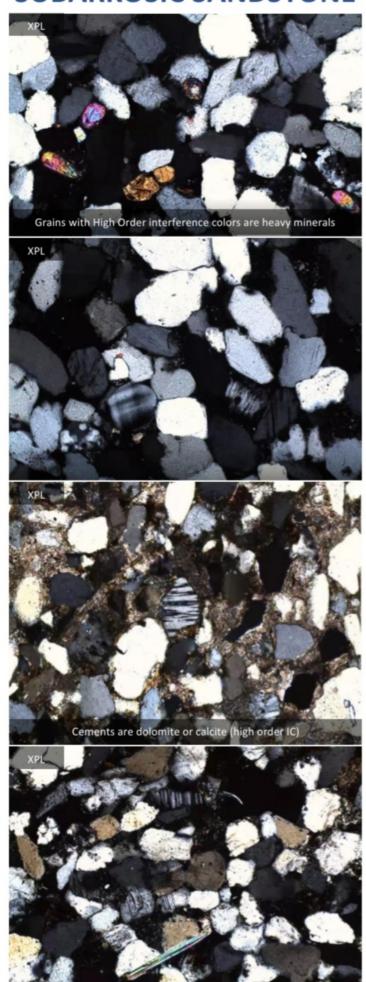


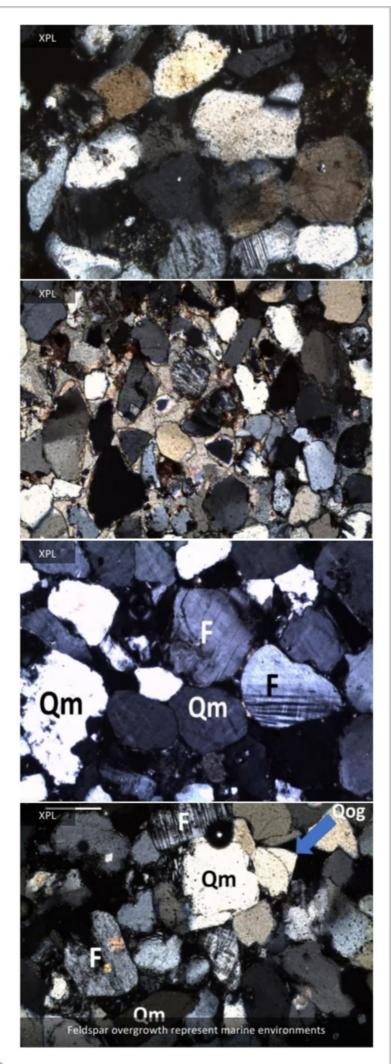
ARKOSIC SANDSTONES





SUBARKOSIC SANDSTONE

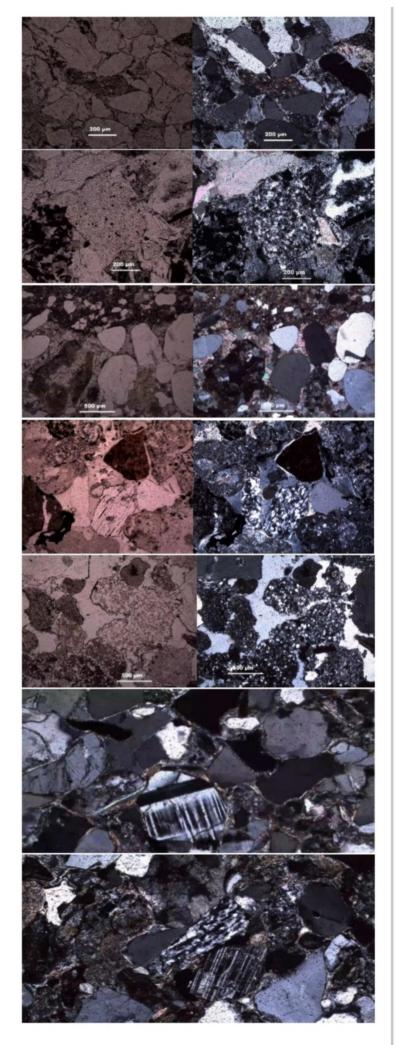


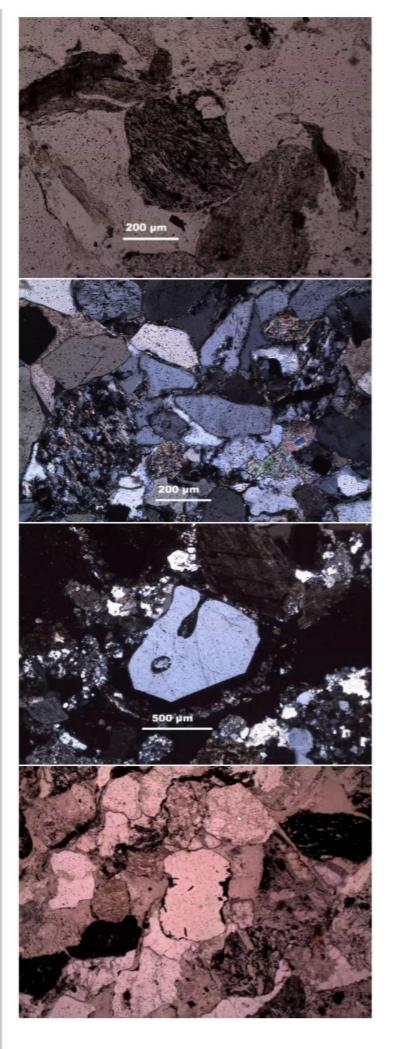


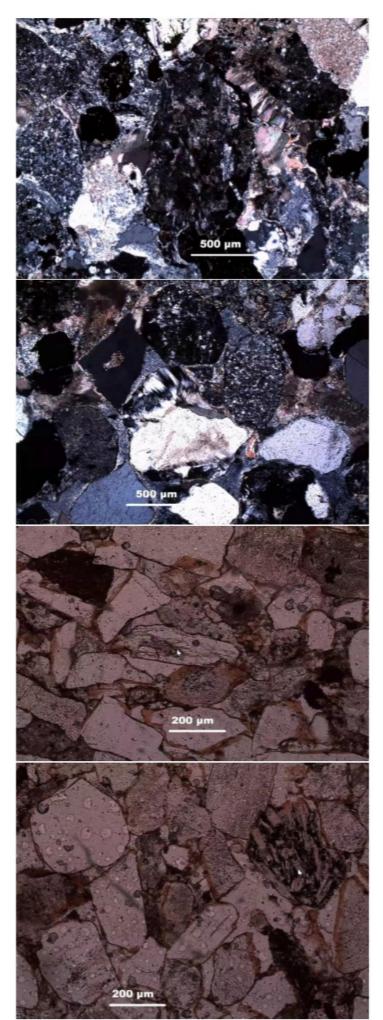
LITHIC SANDSTONES

- Lithic fragments with parallel arranged micas or with polycrystalline quartz have a metamorphic origin
- Rock fragments with feldspar laths have basaltic origin
- Lithic fragments with silt-sized quartz grains have a sedimentary origin (siltstones)
- · Lithic fragments with obsidian have volcanic origin
- Fagments with mica or biotite flakes have mafic origin









Carbonates

- Form 10-15 % of the sedimentary rocks
- Limestones are more abundant than dolostones
- Carbonate rocks are normally free of impurities, & contains < 5% clays & fine-grained quartz
- Limestones are recognized in the field by its relative softness & by reactivity with diluted HCl

 $CaCO_3 + 2HCI \rightarrow Ca^{2+} + 2CI^- + CO_2 + H_2O$

- · Dolostone reacts visibly with HCl when powdered
- Dolostones commonly weather to dull brownish material due to the presence of some iron
- The textures of limestone are quite variable due to the complex origins of these rocks
 - Form textures of detrital rock, chemical precipitate,
 & characteristic of growth habits of organisms
- Formed biochemically, or diagenesis (recrystallization, compaction, & cementation at ≥ 200°C)
- Mineralogy
 - ➤ Calcite CaCO₃ (Rhombohedral) Low-Mg (<4Mg, stable), & High-Mg (>4%)
 - Aragonite CaCO₃ (Orthorhombic)
 - Dolomite CaMg(CO₃)₂

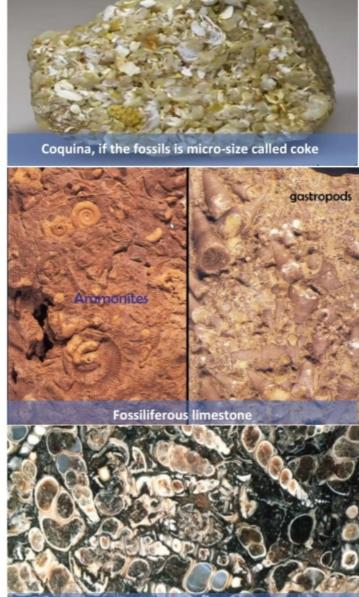
Grains (allochemical, allochem)

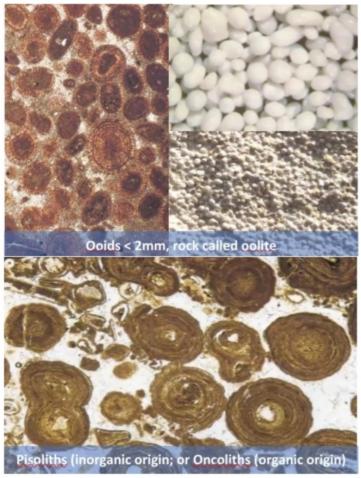
- gravel-, sand-, & coarse silt-size carbonate particles >30 microns that form the framework in mechanically deposited limestone
- 4 grain type: Fossils, ooids, peloids, & limeclasts
 Fossils or skeletal grains: include pelecypods, brachiopods,
 Gastropods, Echinoids, Ostracods, & Corals



- Qoids (<2mm): Spherical, polycrystalline carbonate grains of sand size, have a concentric or radial structure
 - have quartz or carbonate fragments as nuclei
 - Oolitic limestone form in agitated shallow marine waters & commonly have cross beds
- Pisoliths have organic origin, & differ from goids by grain size (larger than goids)







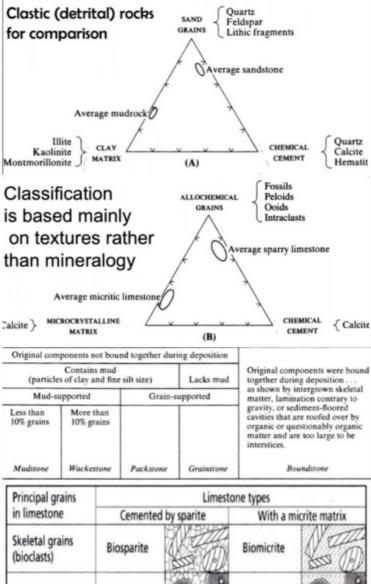
- Peloids sand-sized of microcrystalline calcite that lacks any of internal structure, elliptical to spherical in shape
 - Represent fecal pellets (contain organic matter)



- Limeclasts: Fragments of earlier formed limestone or partially lithified carbonate sediment
 - 2 types of limeclasts: intraclasts & Extraclasts, Most are intraclasts (pieces of penecontemporaneous partially lithified carbonate sediment from within the basin of deposition)

Limestone Classification

- The matrix in limestones
 - mud (micrite): Ca-carbonate mud (1-5μm), binds allochemical grains, & represents low energy env.
 - 2. Microspar 5-15 µm, recrystallization of micrite
 - 3. Sparry calcite cement > 20 microns
- Insoluble residues: chert, clay, detrital quartz
 - shells of radiolarian & diatom (consist of silica)
 - The silica is present in the form of nodular form parallel to limestone bedding planes



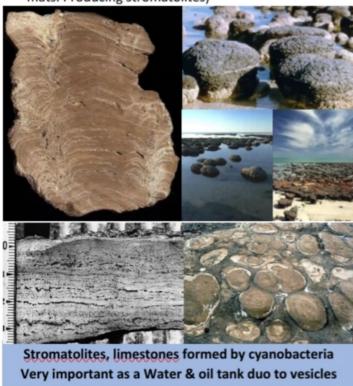
Principal grains	Limestone types							
in limestone	Cemented	by sparite	With a micrite matrix					
Skeletal grains (bioclasts)	Biosparite		Biomicrite	1 222 (
Ooids	Oosparite		Oomicrite	0				
Peloids	Pelsparite	0,9	Pelmicrite	0,0				
Intraclasts	Intrasparite		Intramicrite	2				
Limestone formed in situ	Biolithite	190	Fenestral limestone- dismicrite	88				

TABLE 16-1 Names and compositions of various limestones

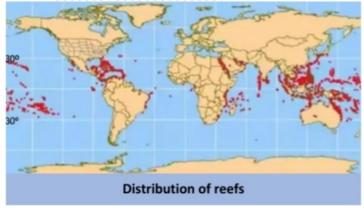
	Composition of limestone					
Allo	chems	Orthochems	Folk name	Dunham name		
70% 30%	Pelecypods Ooids	Sparry cement	Oolitic pelecypod biosparite	Oolitic pelecypod grainstone		
50% 5% 15%	Ooids Fossil fragments Glauconite	Sparry cement	Glauconitic oosparite ⁿ	Glauconitic oolitic grainstone		
÷0%	Peloids	Microcrystalline carbonate matrix	Fossiliferous pelmicrite	Possiliferous peloidal wackestone or packstone		
30% 10%	Fossil fragments Intraclasts					
70%	Intraclasts (gravel size)	Sparry cement	${\bf Trilobite\ intrasparudite}^b$	Intraclastic grainstone		
25% 5%	Trilobites Peloids					
40% 40% 20%	Crinoid fragments Brachiopods Clay minerals	Microcrystalline carbonate matrix	Clayey crinoid-brachiopod biomicrite	Crinoid-brachiopod wackestone or packstone		

Structure

- All structures of sandstones that generated by currents can forme in limestones: e.g. cross bedding,imbrication
- Geopetal sructuires The shells of organisms aiter their death settie down with concave side pointing downwar, this used to determine up direction of the beds
- Lamination: organic or inorganic, & the most important type are produced by soft- bodied, filamentous blue green algae that grow as mats (mats are swept over by wave fine carbonate mud becomes attached to these mats. Producing stromatolites)

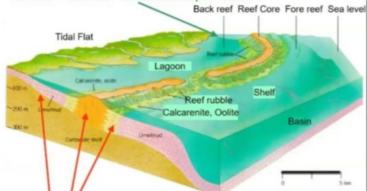


- Micrtic Mud Mounds: Circular to oval dome-shaped accumulations of microcrystalline calcite
 - Thicknesses up to 100's m & diameters up to 1km
 - > They can form shallow & deep environments
 - Reef formed from accumulation of mud mounds
- Organic reefy Carbonate buildups of local origin that are laterally restricted
 - Reefs originate almost entirely in low latitudes in shallow marine waters
 - The carbonate sediments are produced by a variety of frame-building organisms: corals, sponges, algae, bryozoas, rudist pelecypods



Example of facies: A reef complex on a carbonate shelf or platform

The depositional environments are: tidal flats - lagoon - back reef - reef core - fore reef - continental slope - basin



The "facies" are the rocks, derived from sediments that were deposited in those environments.

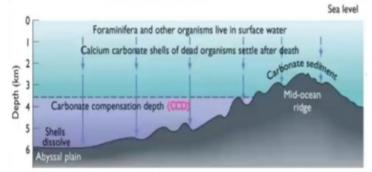




Calcium Carbonate Depositional sites

- Warm T necessary to cause supersaturation of waters with respect to CaCO₃ (abundance & growth of calcareous-shelled organism)
- Calcareous-shelled organism are mostly marine & need light, constant salinity, clear & warm waters, Water must be very shallow & far from large rivers which cause a drop in salinity
- Modern reef-building organisms (corals) contain bluegreen algae which needs light to thrive
- Carbonates Compensation level (CCL) where calcium carbonate breaking down deeper in equatorial regions because carbonate formation is larger in warmer water
 - > CCL pelagic carbonate, 4km in open ocean (photic)
 - ➤ Under CCL, T decreases so solubility of CaCO₃ increases to produce H₂CO₃ which controlling the solubility of limestone (unstable in acidic env.)
 - PH on the surface (7.9 8.1) & as the concentration of H₂CO₃ increases pH became (7.8) limestone at this pH disintegrates (limestone fense: pH = 7.8)

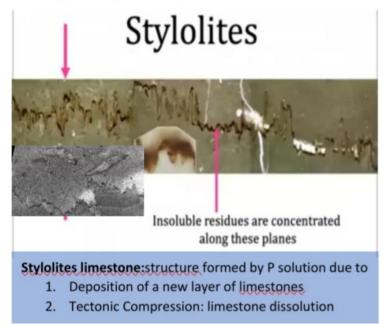
- The limestone can be formed over the bridge because it's a high area & under CCL it melts & chert is more stable (cryptocrystalline quartz)
- chert is formed as a result of the sink of the shells to the oceanic floor, which build their shells from Qz or CaCO₃, whose shells are built from Qz (e.g. idolaria & diatomes) forms pelagic silica (chert, melts 8.5pH)
- Carbonate secreting planktonics didn't evolve until Jurassic, microbiomicrite of pre-Jurassic didn't exist



- Lacustrine Carbonate: lake deposits & commonly associated with other evaporites
 - During late spring & early summer surface waters of lakes turn white as T increase, removal of CO₂ from surface waters is at maximum as a result of active photosynthesis by microscopic plant "charophyte"

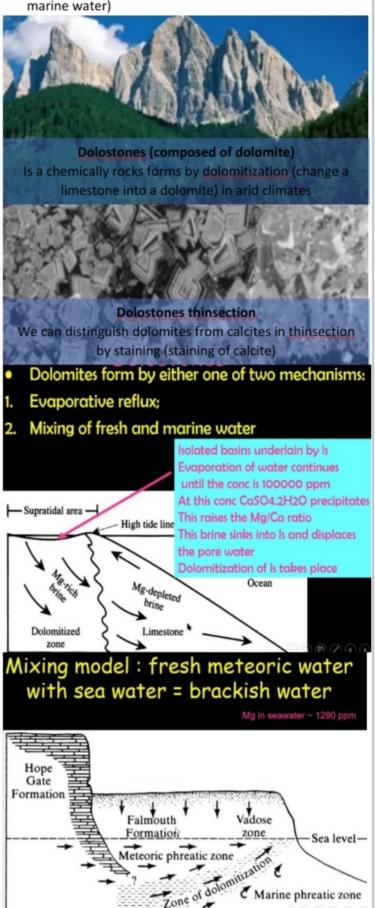
8.8 IAP & Diagenesis

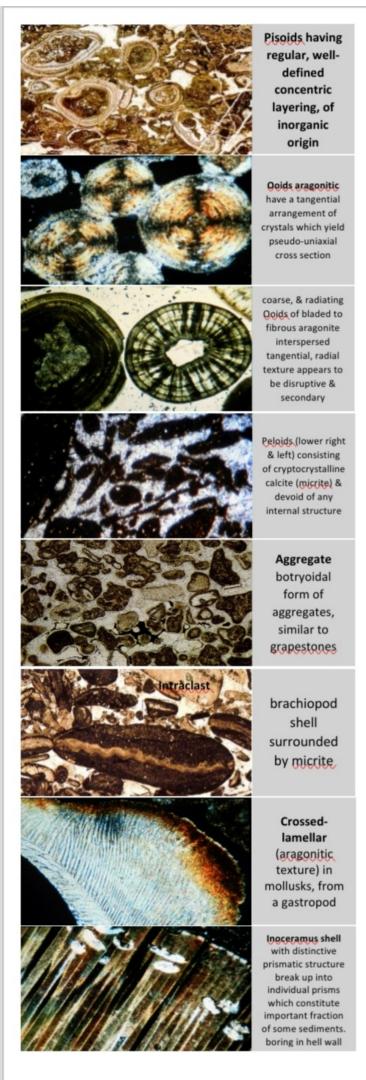
- Diagenesis: all process that contributes to solidification of sediments into a sedimentary rock include cementation, compaction, Lithification
 - start after deposition of skeletal carbonates
 - Cementation by production of hardground on the shelf carbonates or beach rocks (Meteoric water cementation)
 - Mechanical compaction & Chemical cementation (including P solution)



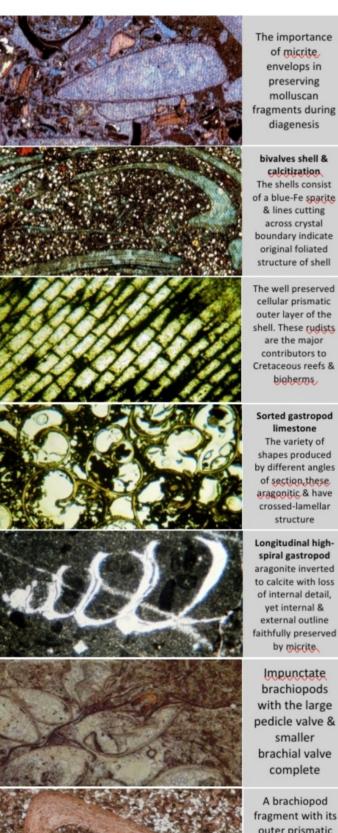
Dolostones

 Dolostones impure carbonate rock formed by one or 2 mechanisms (Evaporative reflux or mixing of a fresh & marine water)







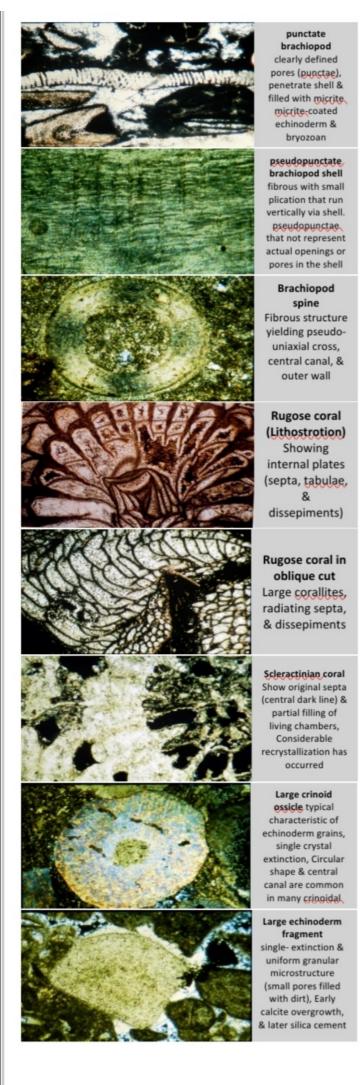


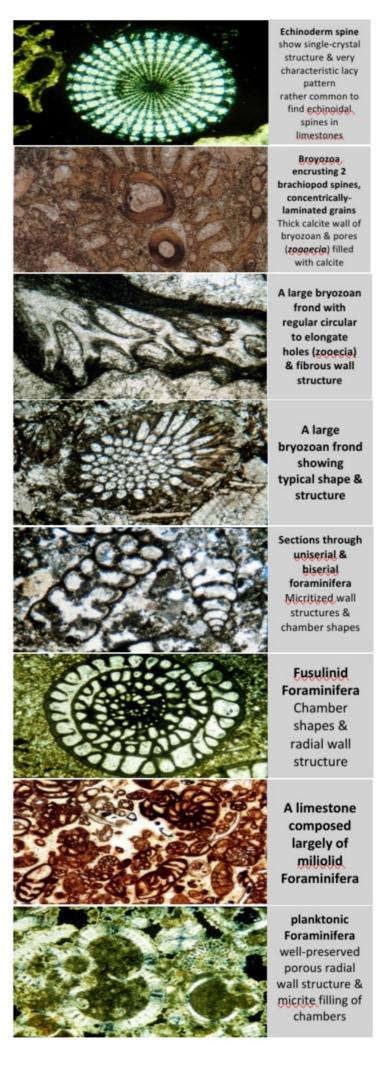
fragment with its outer prismatic layer preserved. The brachiopod

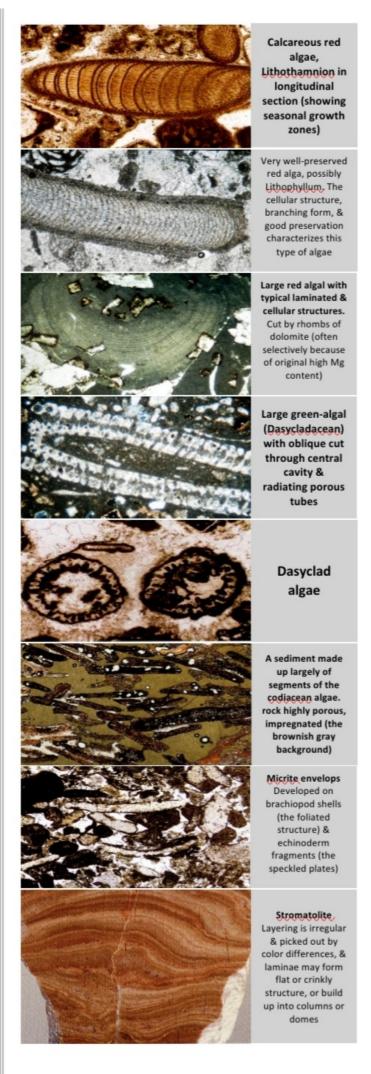
punctae **Broken brachiopod** consist of 2 valves, & micrite envelope Fibrous structure & the fine tubes at right angles to the wall (endopunctae) sparis blue stained due to high Fe

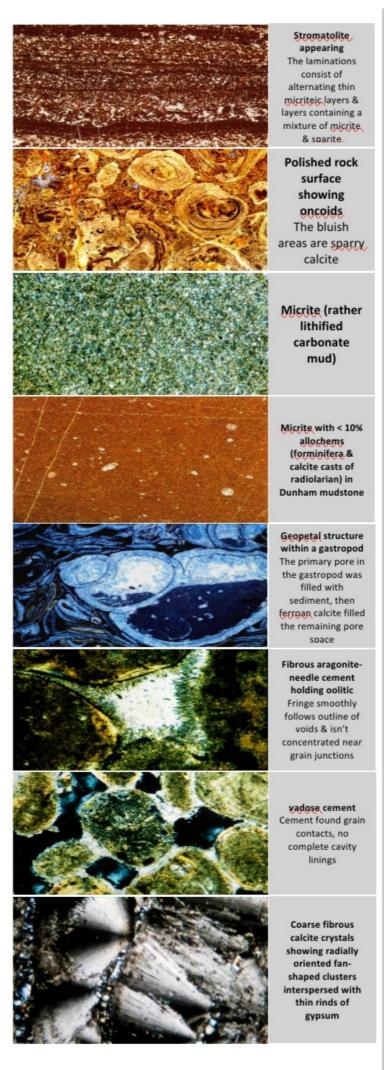
is impunctate

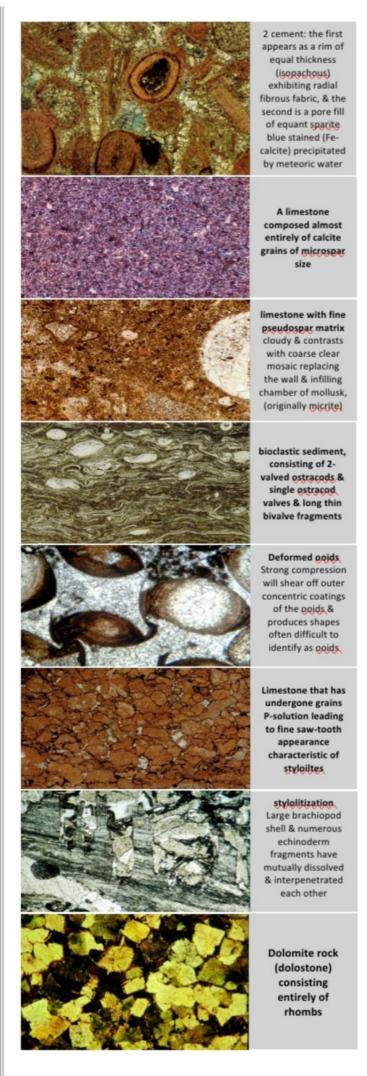
having no

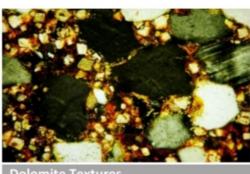












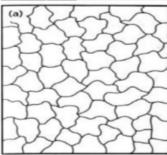
Dolomite rhombs replacing a micritic matrix between framework Qz & Fs in a sandstone Micritic of reddish color not dolomitized & remaining between particles

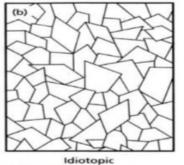
Dolomite Textures

Xenotopic

Anhedral crystals with curved to serrated & irregular crystal boundary

Euhedral rhombic crystals







Dolomite crystals are commonly zoned, the inner part is more cloudy (from fluid inclusions or calcite relics) & the outer part is clear

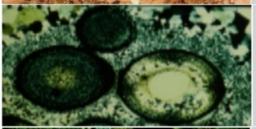


Poikilotopic dolomite engulfing several Qz & Fs in an arkosic arenite sandstone

The dolomite is stained turquoise due to Fe-content



Dedolomite consisting of calcite spar that exhibits rhombs outlined by Fe-oxide & rich in inclusions indicative of former dolomite rhombs



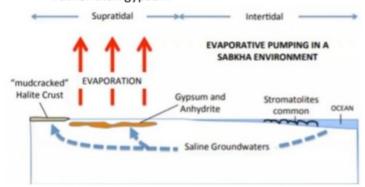
Silica replacement of carbonate (oolitic) sediment Oz overgrowths on nuclei of ooids, chert replacement of ooids



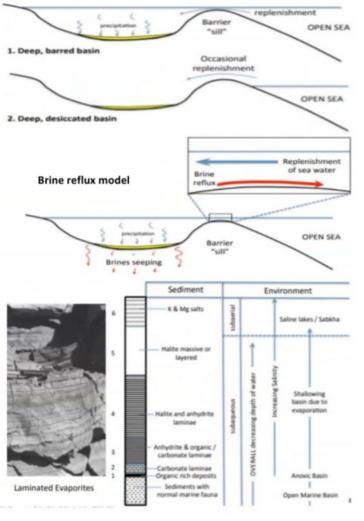
Chalcedonic Qz in a calcareous sponge The well-developed radiating bundles of chalcedony fibers cut by growth banding, & the growth interference between chalcedony bundles

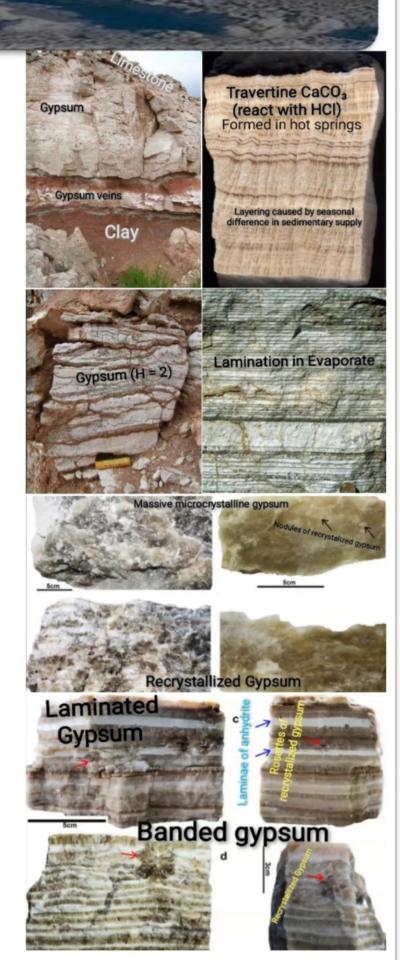
Evaporites

- Sabkha Supratidal environments along arid coastlines build out 'prograde' OVER intertidal sediments giving them a unique sequence of facies in sedimentary logs.
 - Supratidal Intertidal Gypsum precipitated directly into sediments (also dolomitization), Gypsum has nodules, chicken-wire, or enterolithic textures
 - May get anhydrite precipitating further landward rather than gypsum



- Deep Basins: Most important source of evaporite
 - 1. Deep basins barred by a barrier: fault,CO₃bank,reef
 - 2. Deep basins with occasional access to sea





















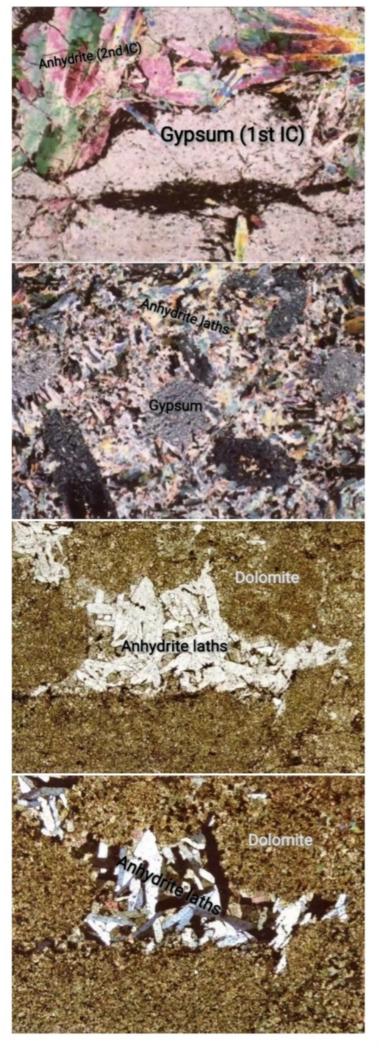






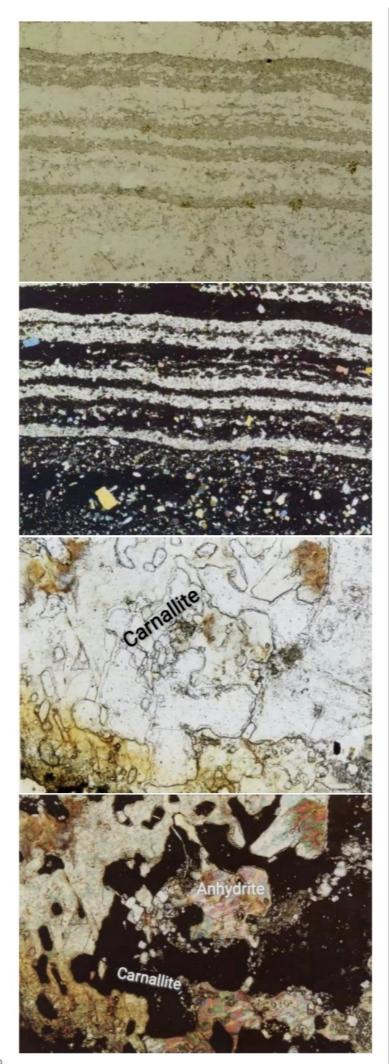


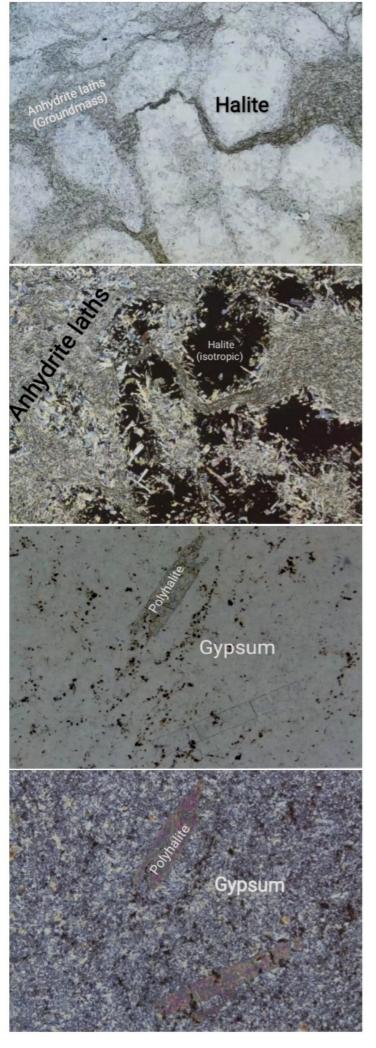


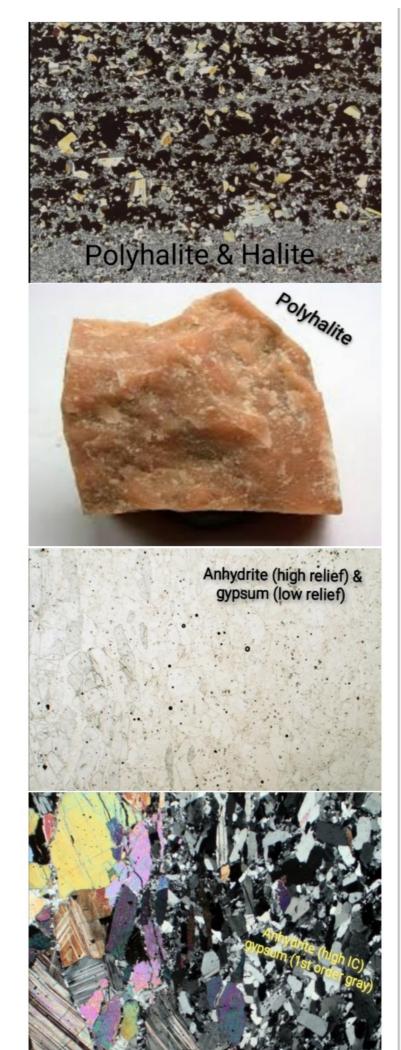




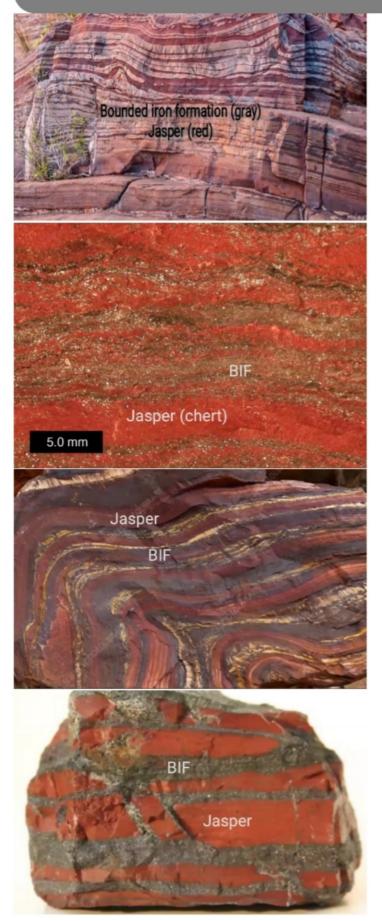


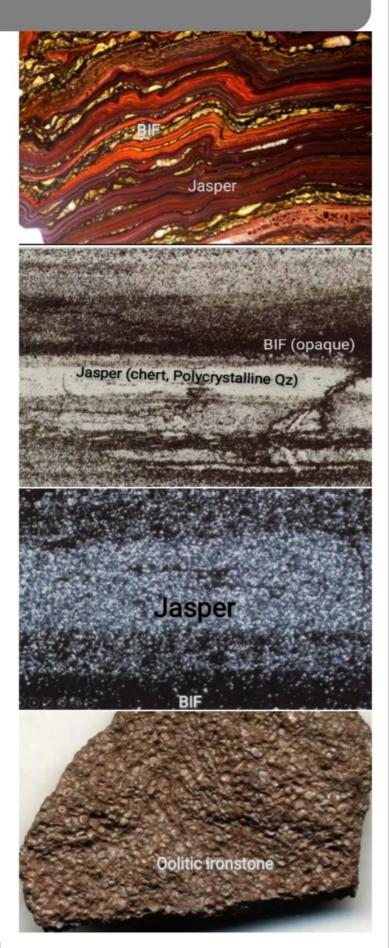


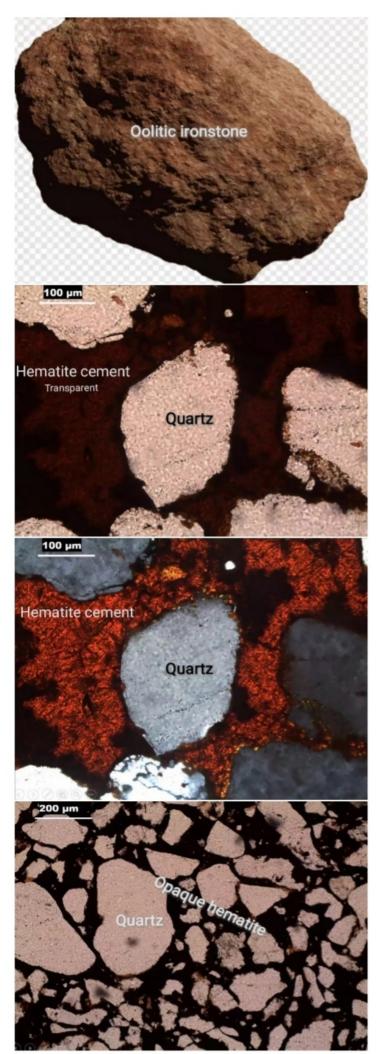


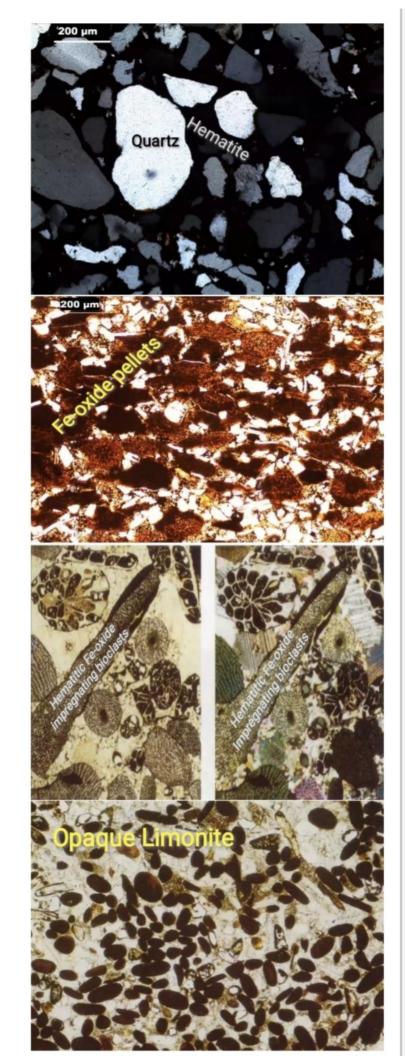


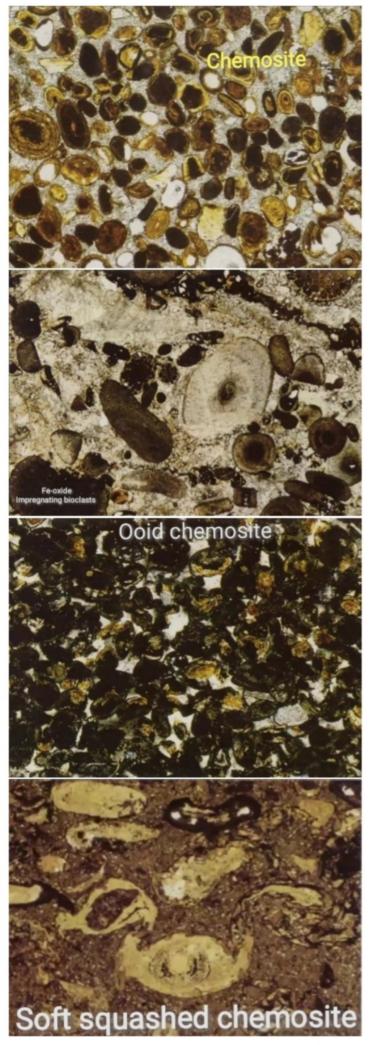
Ironstones

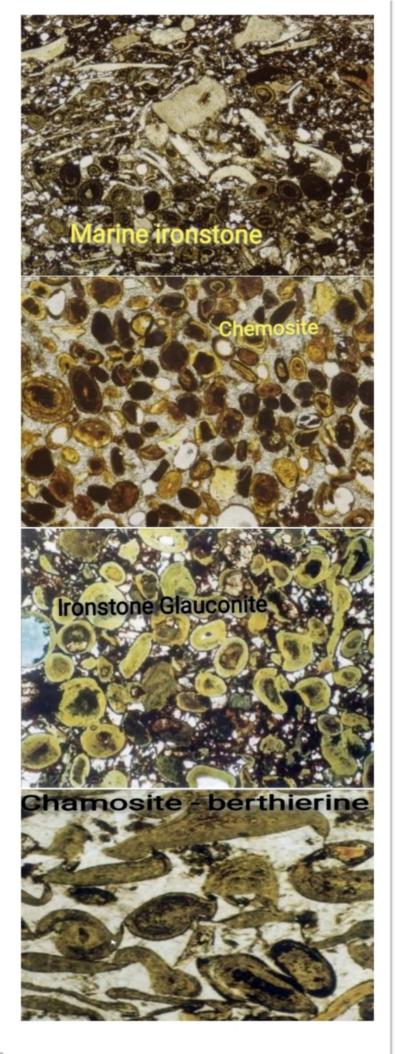


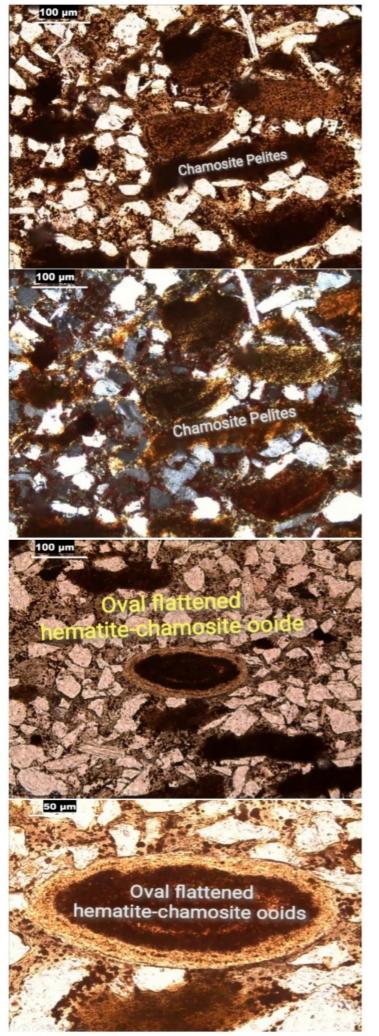


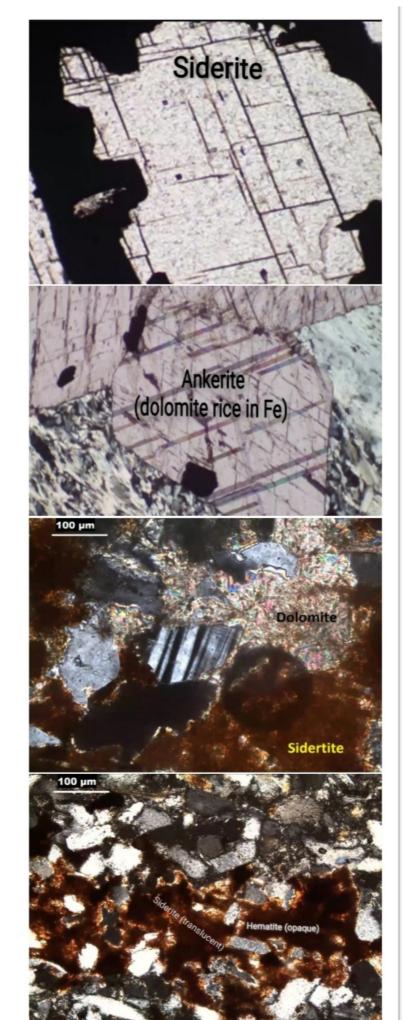


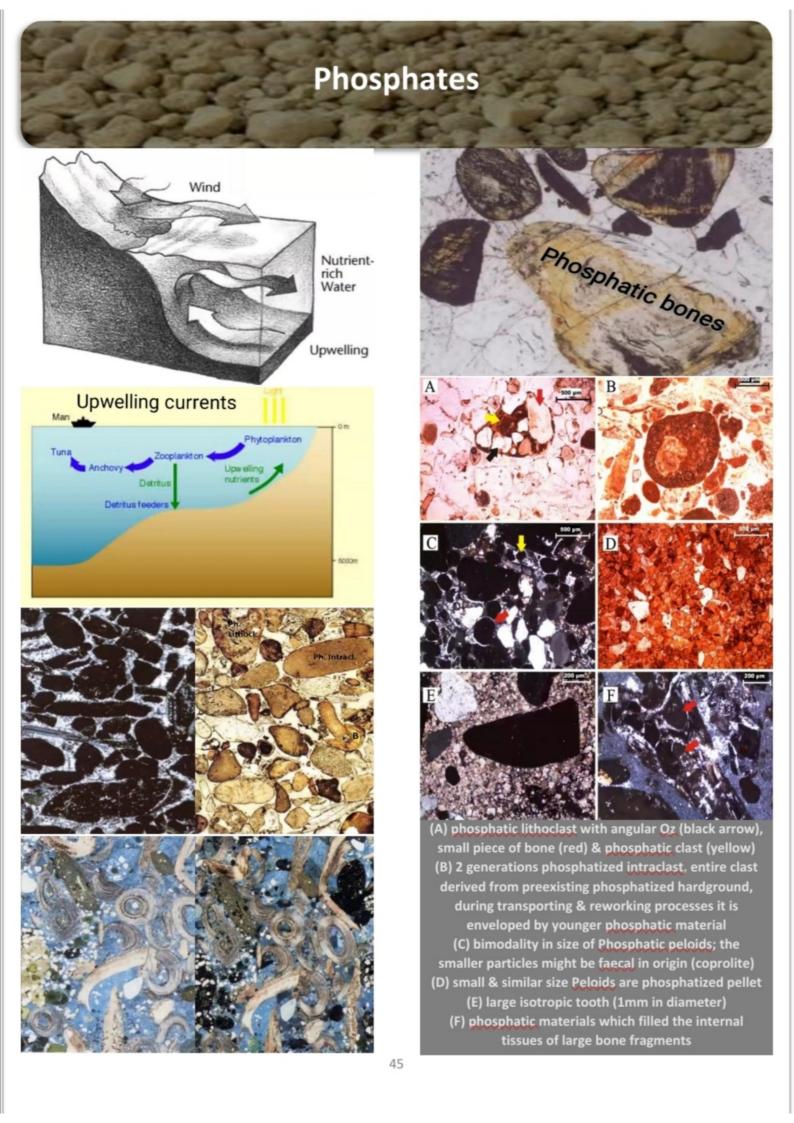


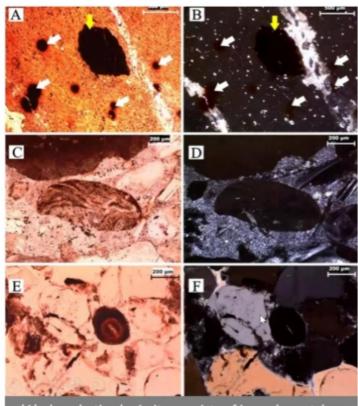






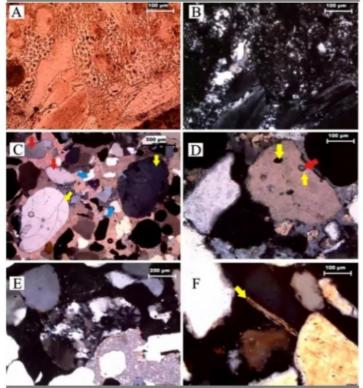




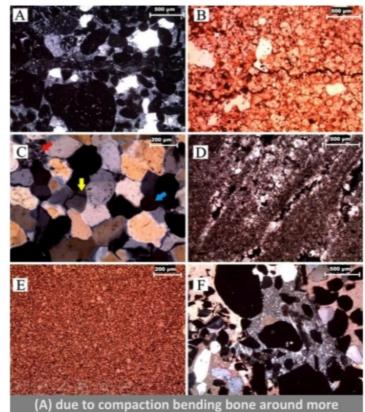


(A) phosphatized micrite consists of irregular patches of organic matter (yellow arrow) & patches of Fe-oxide (B) phosphatized micrite

- (C) phosphatized shell
- (D) phosphatized shell floating in cryptocrystalline Oz
 (E) Phosphatized oolite
- (F) Phosphatized oolites surrounded by silica cement



(A + B) phosphatized algae
(C) monocrystalline Oz cemented by calcite
(D) The abraded Oz includes heavy minerals (yellow arrows) & fluid inclusions (red arrow)
(E) Polycrystalline Oz with sutured boundaries
(F) Micaceous flake embedded in monocrystalline Oz



rigid quartz grains & phosphate clasts is seen

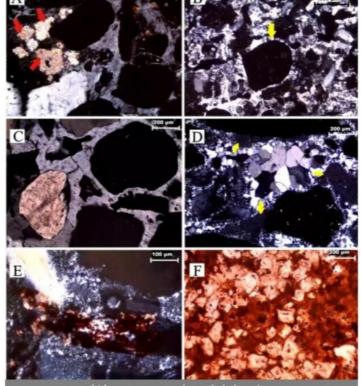
(B) μ-stylolite structure (Fe-oxide solutions pathways)

(C) Oz syntaxial overgrowth, rutile, & Feldspar grain

(D) mulloscs shells in micritic ground

(E) Mosaicing dolomite

(F) calcite poikilitic cement



(A) gypsum replaced dolomite

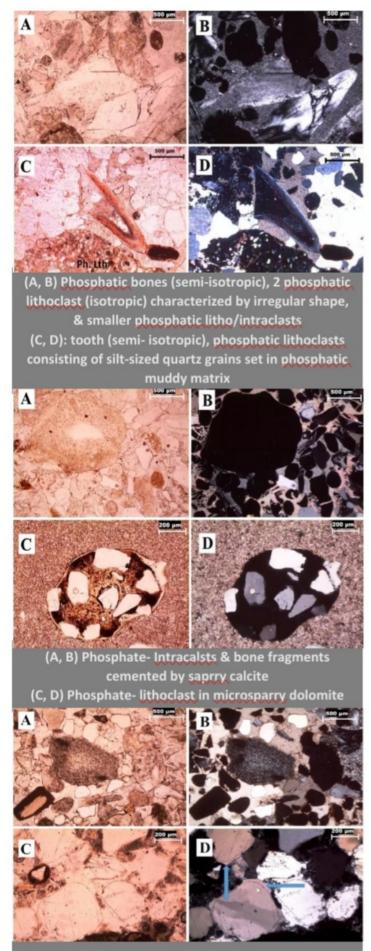
(B) isopachous gypsum cement phosphatic clast

(C) poikilitic texture of gypsum cement

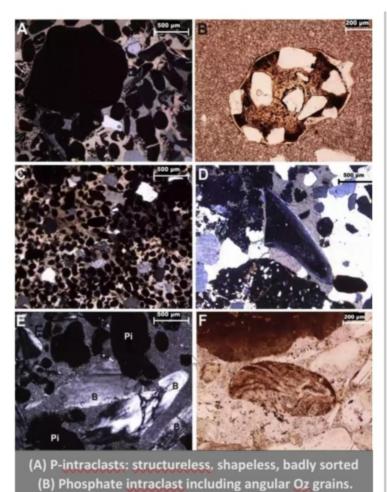
(D)Yellow arrows assign micrite patches

(E) Iron oxides replaced piece of bone partially

(F) Organic matter replaced dolomite



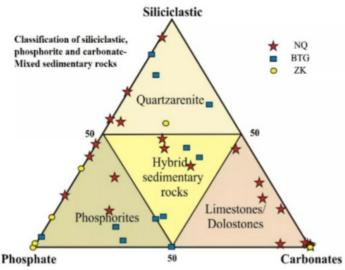
(A, B) 2 large lithoclasts that are angular to subangular (silicified, & phosphatic) with smaller phosphatic intraclasts & detrital quartz grains
 (C, D) Detrital monocrystalline quartz grains exhibiting syntaxial quartz overgrowth



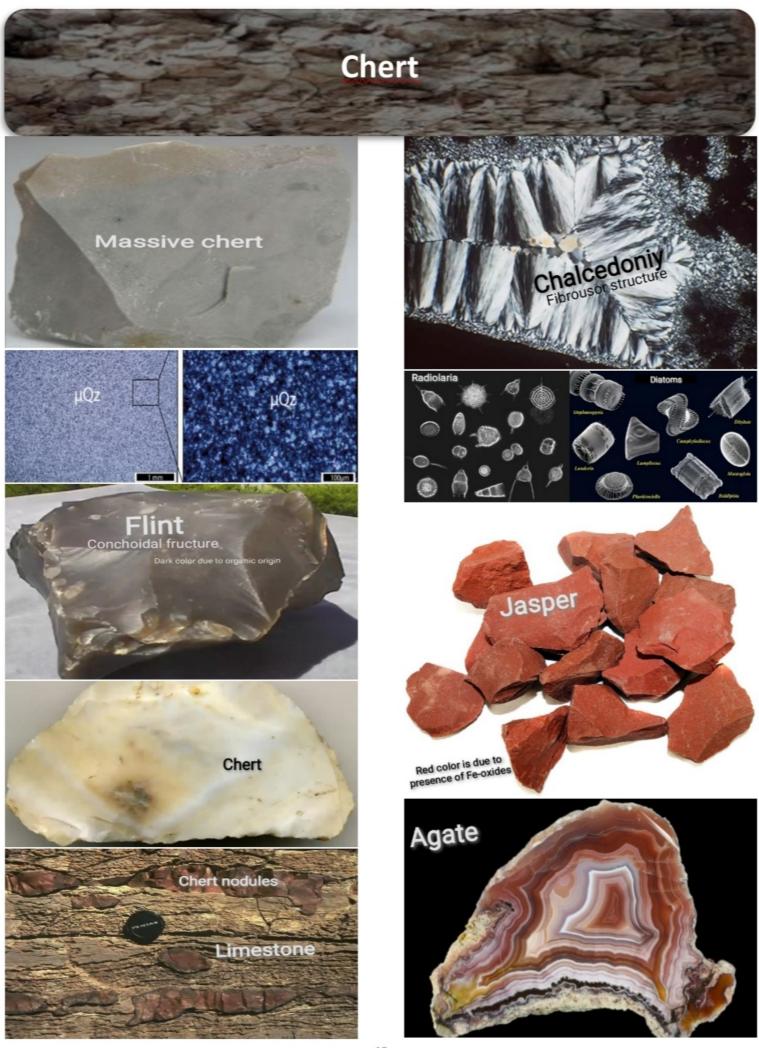
Jordan phosphates

bones show 1st order grey

(C) Phosphate peloids in calcite poikilotopic cement
(D) Large isotropic (1st grey) tooth, 1mm in diameter
(E) Bone fragments (B) & phosphate intraclasts (Pi). The



No.	Silic.	Phos.	Car.	Rock name
NQ9	39.4	60.1	0.50	Arenaceous phosphate grainstone
NQ10	22.6	77.4	0.00	Siliceous phosphate grainstone
NQ16	44.7	55.3	0.00	Arenaceous phosphate packstone
NQ18	30.0	55.3	14.7	Dolomitic Siliceous P-Packstone
NQ19	15.3	64.9	19.8	Arenaceous Dolomitic P-Packstone
BTG11	11.8	48.2	36.1	Arenaceous calcareous Phosphate Packstone
BTG12	13.4	46.4	40.2	
BTG15	8.90	66.0	25.2	
BTG16	4.30	95.7	0.00	Phosphate packstone
BTG18				Siliceous calcareous

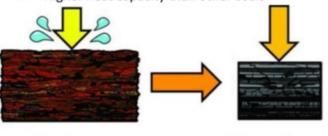




- Formation coals involved 4 stages: peat, Lignite, bituminous, & Finally Anthracite
- Peat & lignite are poor in quality (less heat capacity) due to lesser amount of carbon in them
 - Lignite has more heat capacity than peat

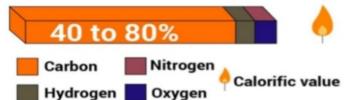


- Lignite transfer into bituminous in the 3rd stage
 - Bituminous: soft & black
 - > Higher heat capacity than other coals



Lignite

Bituminous



- At the last stage of coals formation we get Anthracite
 - Anthracite is a metamorphic coal has shiny luster & very hard, due to it's strongest chemical bonds cannot used as fuel

